

A Decomposition of Droplet Simulation Using Molecular Dynamics

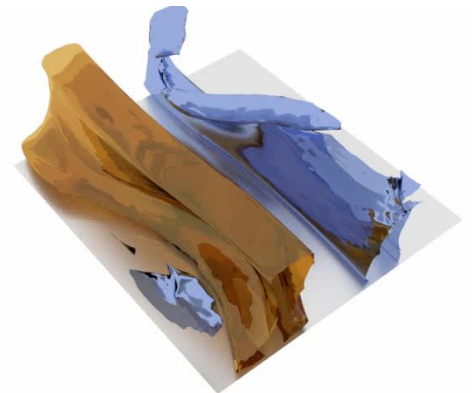
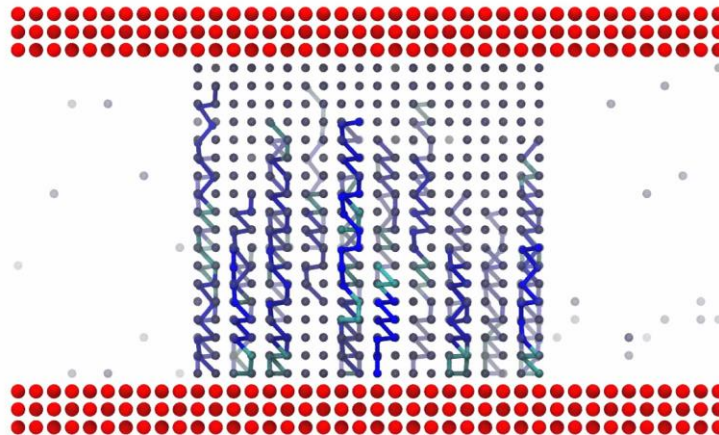
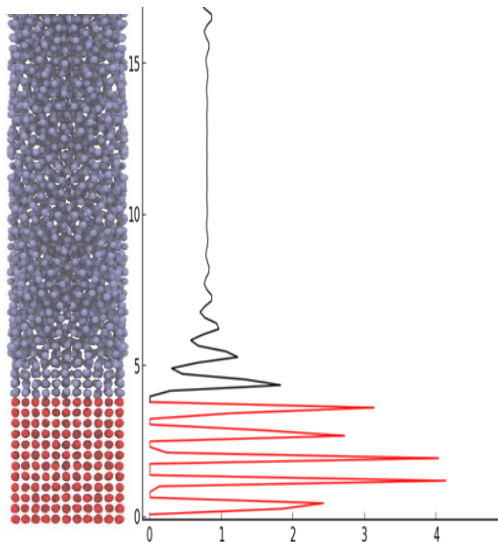
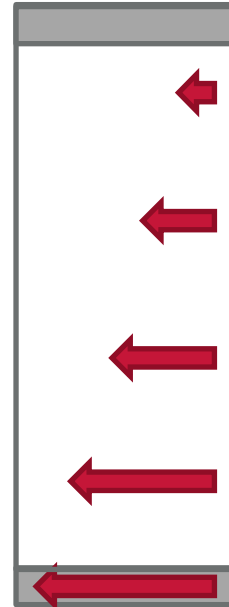
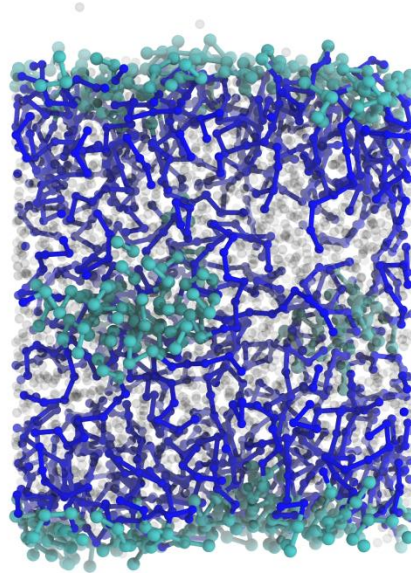
Edward Smith,

Panos Theodorakis, Richard Craster and Omar Matar

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18/09/19

Molecular Dynamics



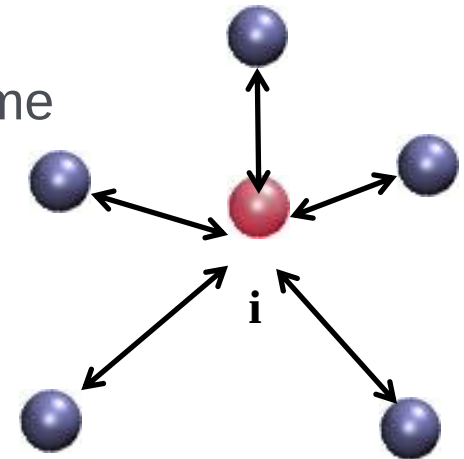
Molecular Dynamics

Discrete molecules in continuous space

- Molecular position evolves continuously in time
- Position and velocity from acceleration

$$\ddot{\mathbf{r}}_i \rightarrow \dot{\mathbf{r}}_i$$

$$\dot{\mathbf{r}}_i \rightarrow \mathbf{r}_i(t)$$



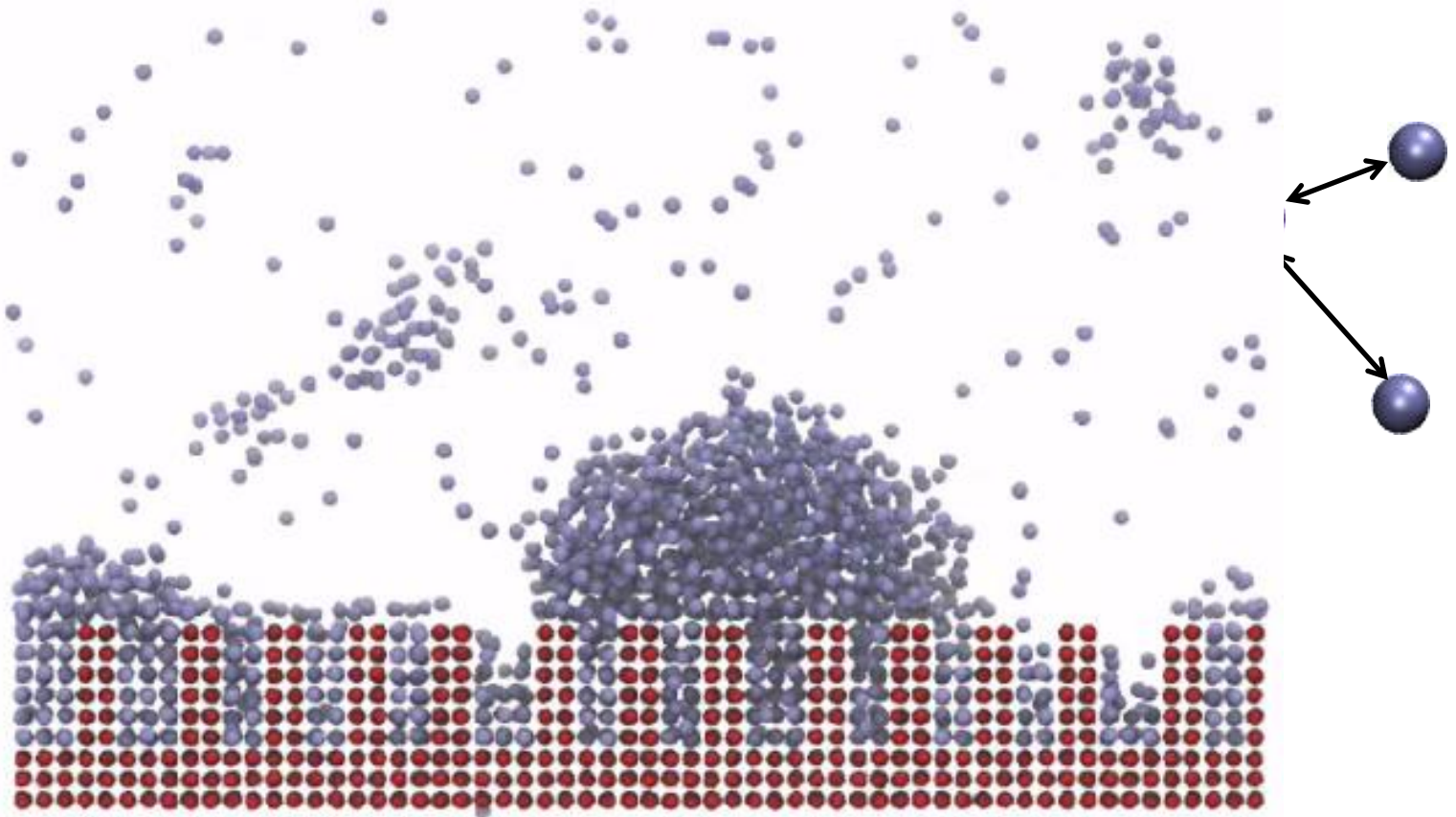
Acceleration obtained from forces

- Governed by Newton's law for an N-body system
- Point particles with electrostatic interactions

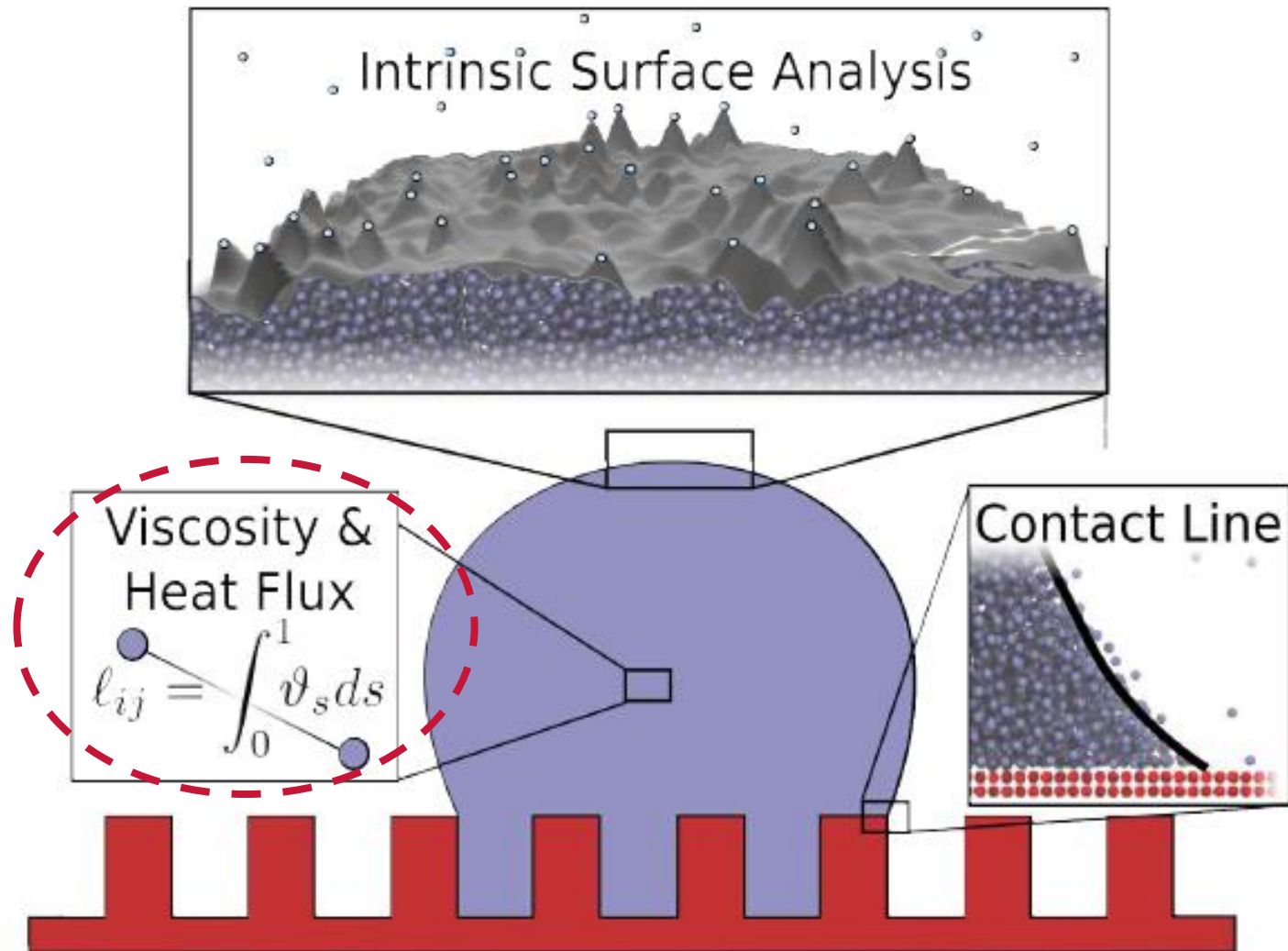
$$m_i \ddot{\mathbf{r}}_i = \mathbf{F}_i = \sum_{i \neq j}^N \mathbf{f}_{ij} = \sum_{i \neq j}^N \nabla \Phi_{ij}$$

$$\Phi_{ij} = 4\epsilon \left[\left(\frac{\sigma}{r_{ij}} \right)^{12} - \left(\frac{\sigma}{r_{ij}} \right)^6 \right]$$

Molecular Dynamics



Overview



Linking Molecular and Continuum

- Assumes continuous fields

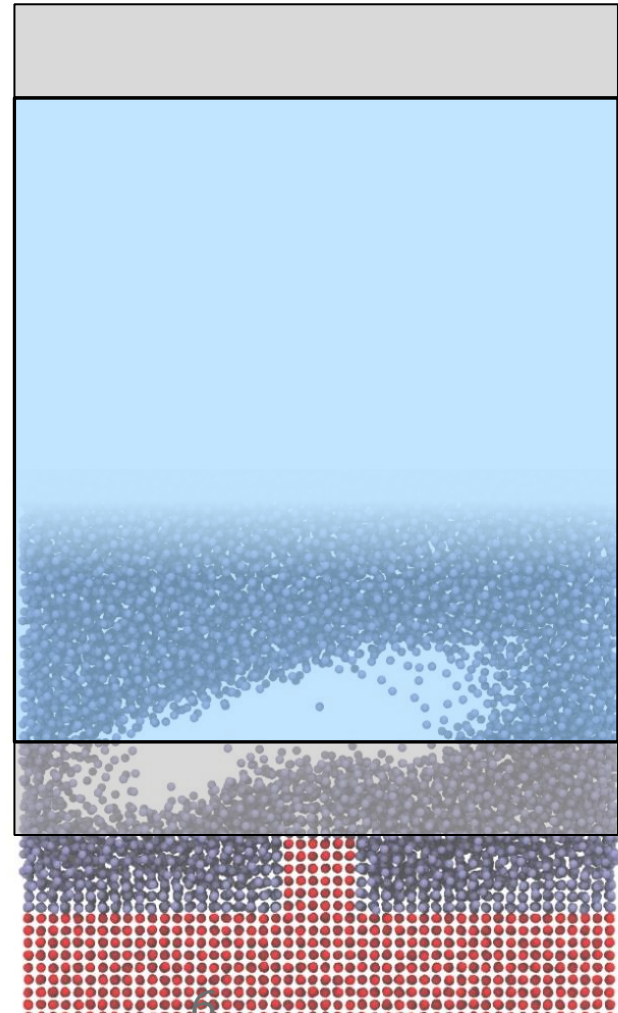
$$\frac{\partial}{\partial t} \rho \mathbf{u} + \nabla \cdot \rho \mathbf{u} \mathbf{u} = \nabla \cdot \mathbf{\Pi} \quad \text{and } \mu, \lambda, \dots$$



How do we get
continuum values
from the molecular
system?

- Discrete molecules

$$m_i \ddot{\mathbf{r}}_i = \mathbf{F}_i \quad \text{for all } i \text{ in } N$$



$$\rho(\mathbf{r}, t) = \sum_{i=1}^N \left\langle m_i \delta(\mathbf{r} - \mathbf{r}_i); f \right\rangle$$

Irving and Kirkwood (1950)

Continuum density
field valid at any
point, $\mathbf{r}$, in space

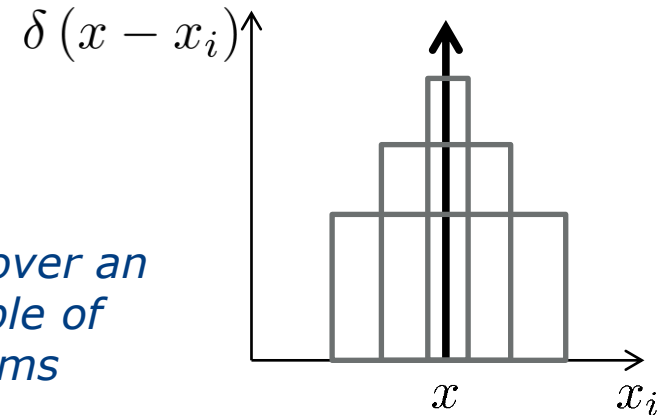
Mass of each
molecule

Average over an
ensemble of
systems

$$\rho(\mathbf{r}, t) = \sum_{i=1}^N \left\langle m_i \delta(\mathbf{r} - \mathbf{r}_i) ; f \right\rangle$$

Sum over all
molecules

Delta function is
only non-zero if
molecule i 's
position $\mathbf{r}_i$ is
exactly at $\mathbf{r}$



The Dirac delta
infinitely high,
infinitely thin peak.
Formally equivalent
to the continuum
differential
formulation

BUT

No molecule is ever
exactly at a point

The Control Volume (Weak) Form

- The “weak formulation” expressed the equations in integrated form

$$\int_V \rho(\mathbf{r}, t) dV = \sum_{i=1}^N m_i \int_V \delta(\mathbf{r} - \mathbf{r}_i) dV$$

- Integrating the Dirac delta function exactly provides a combination of Heaviside functions
 - » Instantaneous description
 - » Direct link between macro and micro description
 - » Exactly conservative

The Control Volume Functional

- The Control volume functional is the formal integral of the Dirac delta functional in 3 dimensions (3D top hat or box car function)

$$\vartheta_i \equiv \int_{x^-}^{x^+} \int_{y^-}^{y^+} \int_{z^-}^{z^+} \delta(x_i - x) \delta(y_i - y) \delta(z_i - z) dx dy dz$$

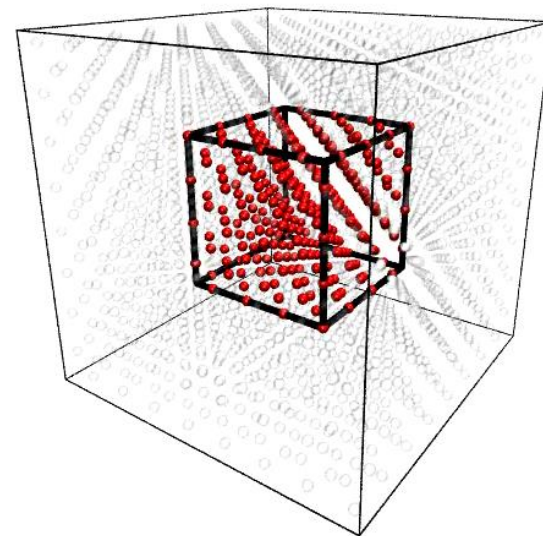
$$= [H(x^+ - x_i) - H(x^- - x_i)]$$

$$\times [H(y^+ - y_i) - H(y^- - y_i)]$$

$$\times [H(z^+ - z_i) - H(z^- - z_i)]$$

- In words

$$\vartheta \equiv \begin{cases} 1 & \text{if molecule is inside volume} \\ 0 & \text{if molecule is outside volume} \end{cases}$$



Derivative yields surface fluxes (Method of Planes)

- Taking the derivative gives flux over the surface of the cube

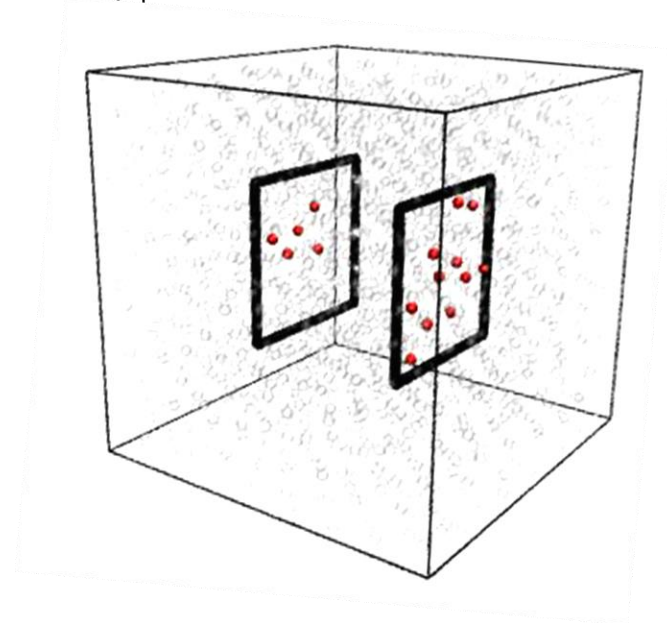
$$dS_{ix} \equiv -\frac{\partial \vartheta_i}{\partial x_i} = [\delta(x^+ - x_i) - \delta(x^- - x_i)] \\ \times [H(y^+ - y_i) - H(y^- - y_i)] \\ \times [H(z^+ - z_i) - H(z^- - z_i)]$$

- Vector form defines six surfaces

$$d\mathbf{S}_i = \mathbf{i}dS_{xi} + \mathbf{j}dS_{yi} + \mathbf{k}dS_{zi}$$

- Or in words

$$d\mathbf{S}_i \equiv \begin{cases} \infty & \text{if molecule on surface} \\ 0 & \text{otherwise} \end{cases}$$

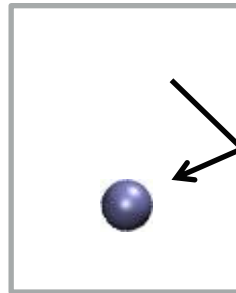


Pressure (Stress) in an MD Simulation

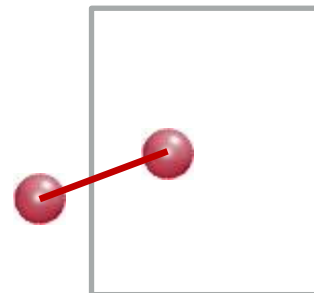
- Pressure definition in a dense molecular system
 - Kinetic part due to fluctuations
 - Configurational part due to liquid structure

$$\Pi_{xy} = \underbrace{\sum_{i=1}^N \left\langle m_i c_{xi} c_{yi} dS_{ix} \right\rangle}_{\text{Kinetic}} + \underbrace{\frac{1}{2} \sum_{i=1}^N \sum_{j \neq i}^N \left\langle f_{xij} r_{yij} dS_{ijx} \right\rangle}_{\text{Configurational}}$$

*Kinetic
theory part
Momentum due
to average of
molecules
crossing a plane
and returning*



$$\mathbf{c}_i = \dot{\mathbf{r}}_i - \mathbf{u}$$



*Configurational
part
Inter-molecular
bonds act like the
stress in a
stretched spring*

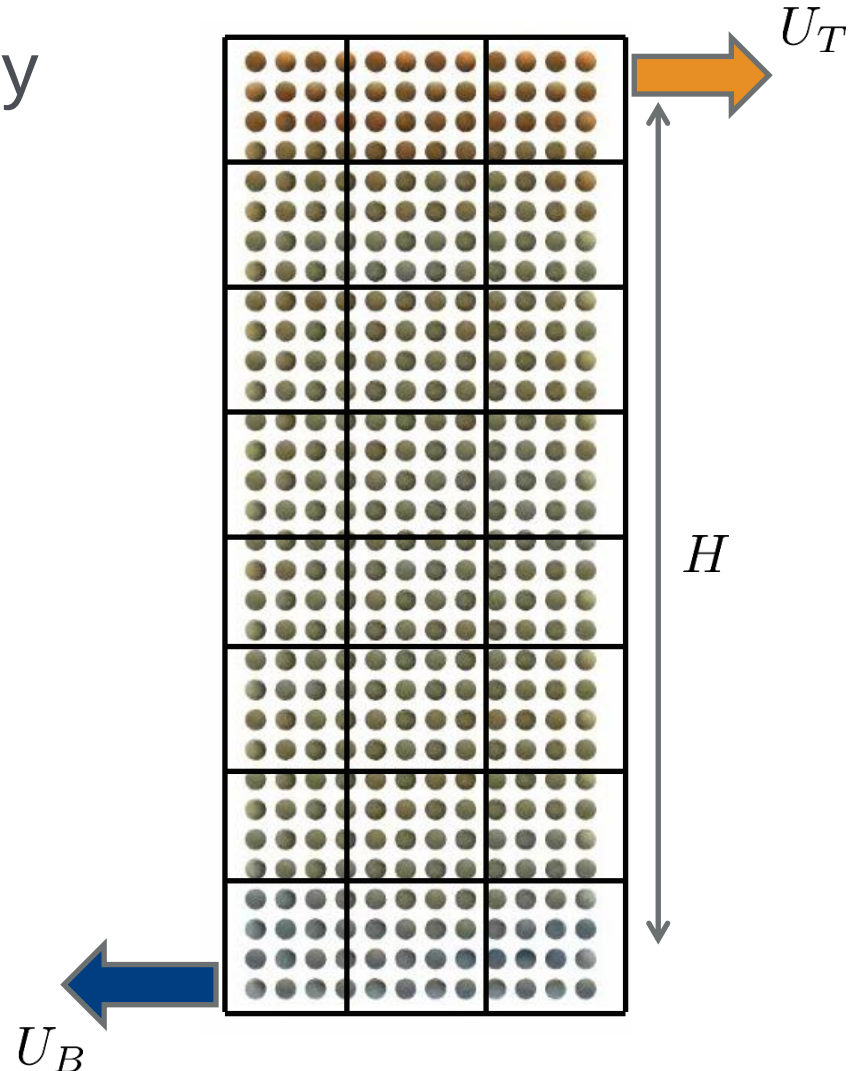
Newton's Law of Viscosity

- Shear flow Π_{xy} driven by sliding walls

$$\frac{\partial u}{\partial y} \approx \frac{U_T - U_B}{H}$$

- Viscosity from stress divided by strain

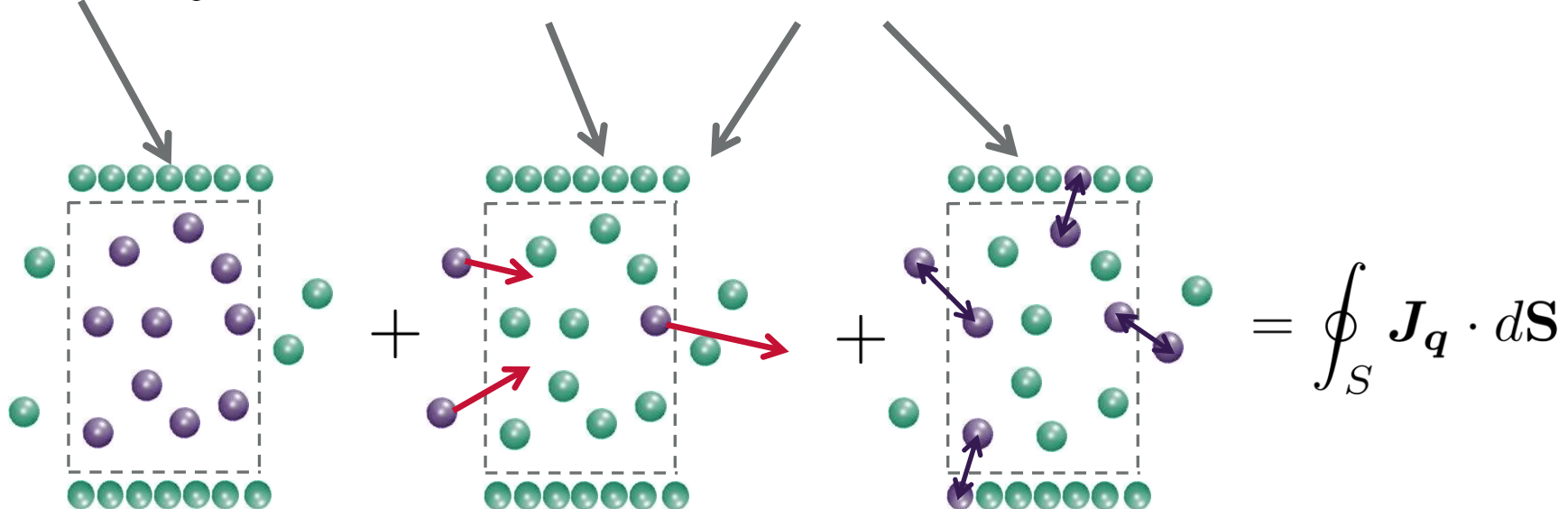
$$\mu = - \frac{\Pi_{xy}}{\partial u / \partial y}$$



Heat Flux in an MD Simulation

- Heat flux is left over after counting all other terms

$$\underbrace{\int_V \frac{\partial}{\partial t} \rho \mathcal{E} dV}_{\text{Unsteady}} + \underbrace{\oint_S \left[\rho \mathcal{E} \mathbf{u} + \underbrace{\boldsymbol{\Pi} \cdot \mathbf{u}}_{\text{Stress Work}} \right] \cdot d\mathbf{S}}_{\text{Advection}} = - \underbrace{\oint_S \mathbf{J}_q \cdot d\mathbf{S}}_{\text{Heat Flux}}$$



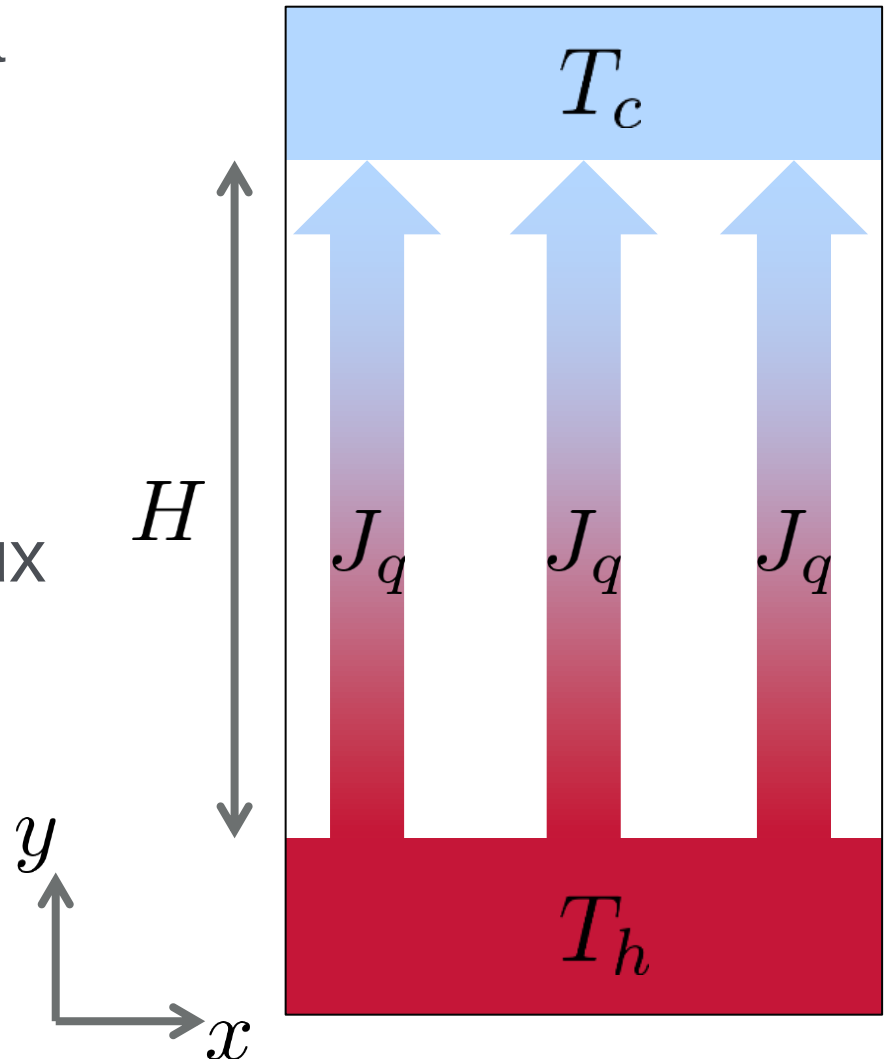
Fourier's law of Heat Conduction

- Heat flux J_q driven by a temperature difference

$$\frac{\partial T}{\partial y} \approx \frac{T_c - T_h}{H}$$

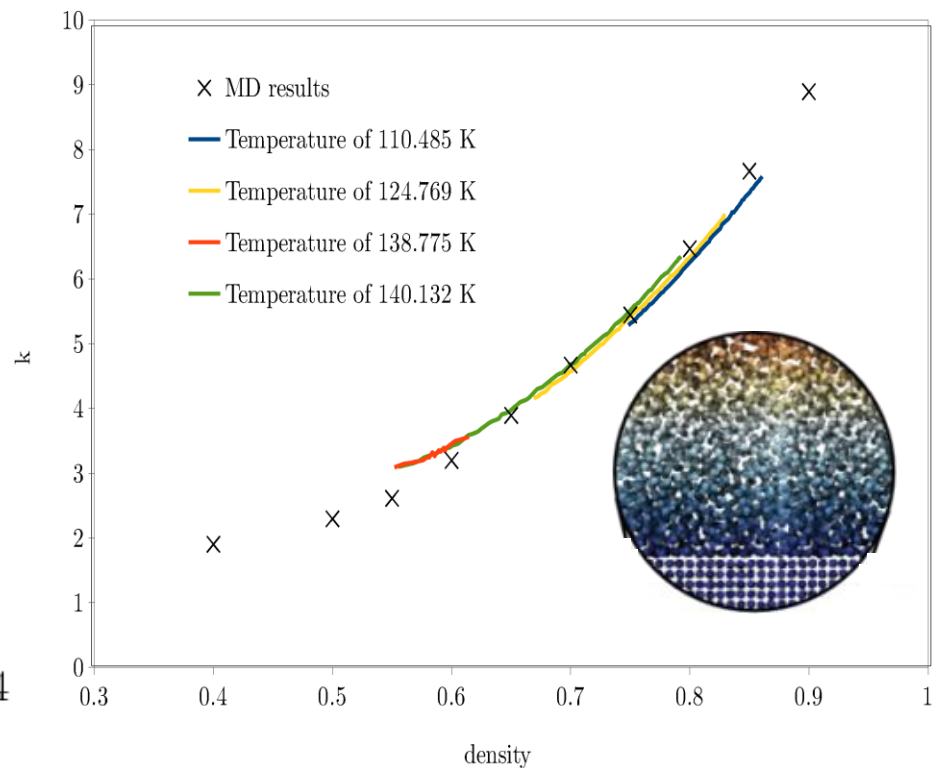
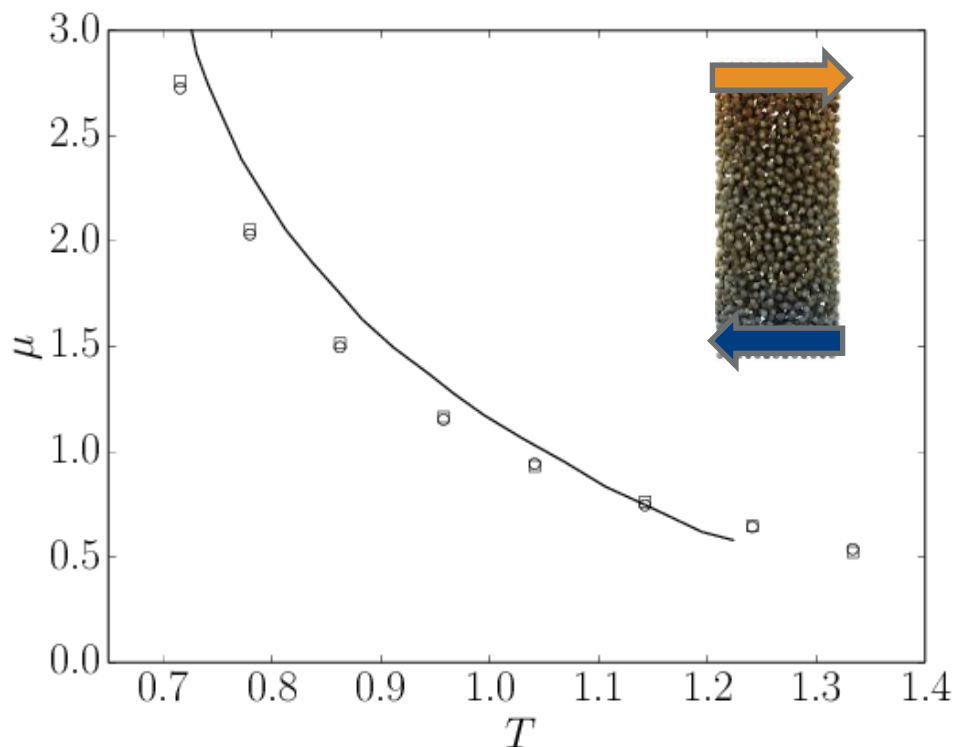
- Coefficient from heat flux divided by temperature gradient

$$\lambda = -\frac{J_q}{\partial T / \partial y}$$

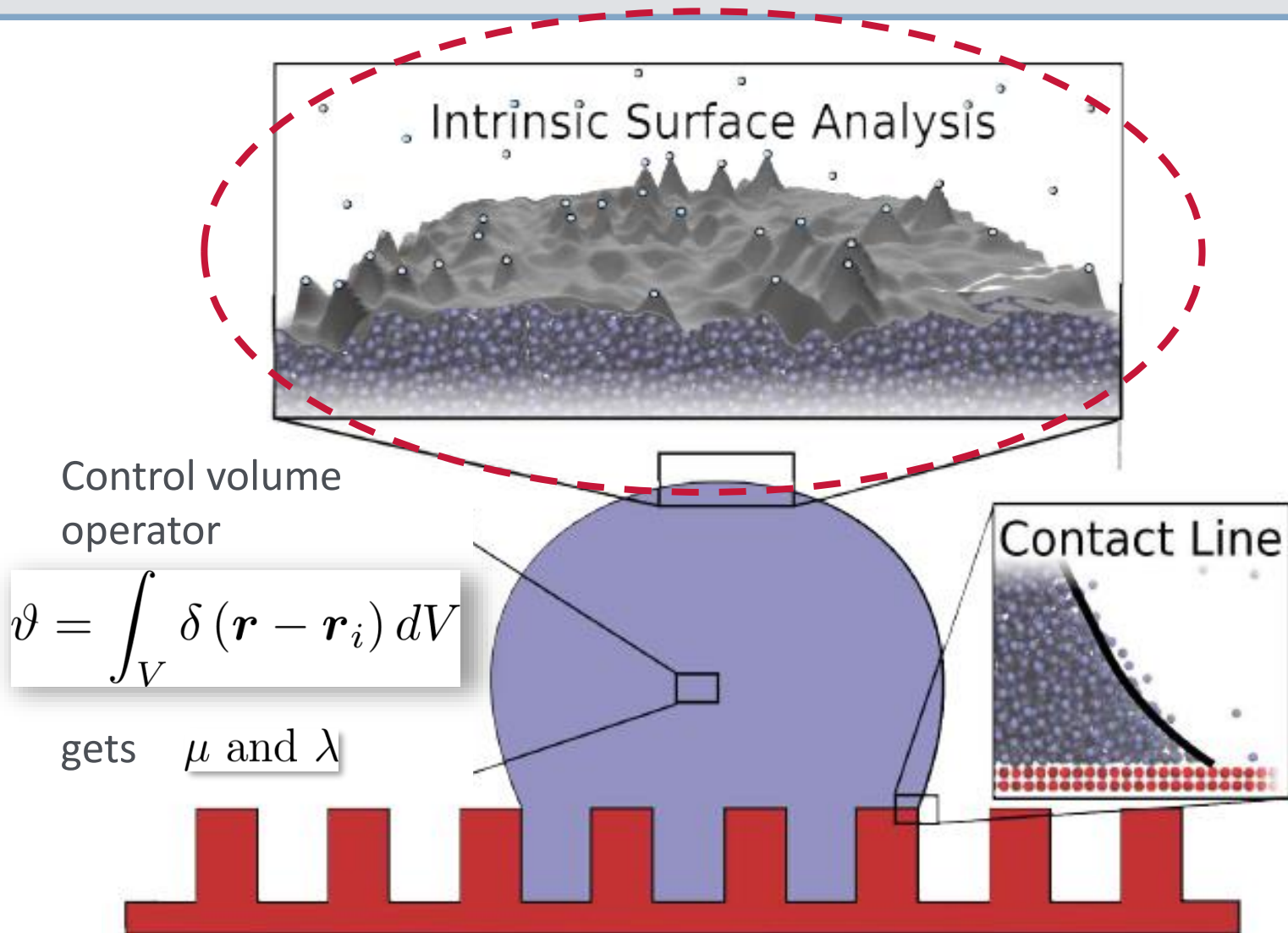


Validation Against Experiments

- Outputs of the simulation match experiments, shown here for liquid Argon



Overview

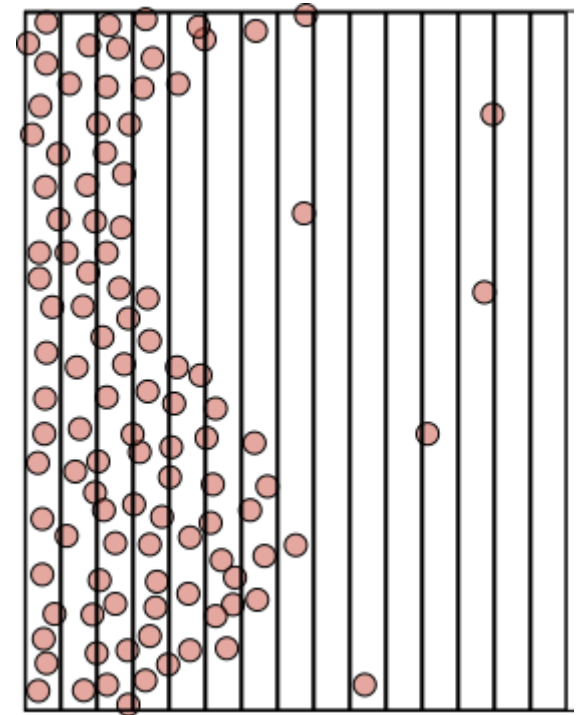


Getting Quantities Near an Interface

- Density in a cubic control volume

$$\begin{aligned}
 \int_V \rho(\mathbf{r}, t) dV &= \sum_{i=1}^N m_i \int_V \delta(\mathbf{r} - \mathbf{r}_i) dV \\
 &= \sum_{i=1}^N m_i \left[H(x^+ - x_i) - H(x^- - x_i) \right] \\
 &\quad \times \left[\cancel{H(y^+ - y_i)} - \cancel{H(y^- - y_i)} \right] \\
 &\quad \times \left[\cancel{H(z^+ - z_i)} - \cancel{H(z^- - z_i)} \right]
 \end{aligned}$$

- Assume a periodic domain in y and z

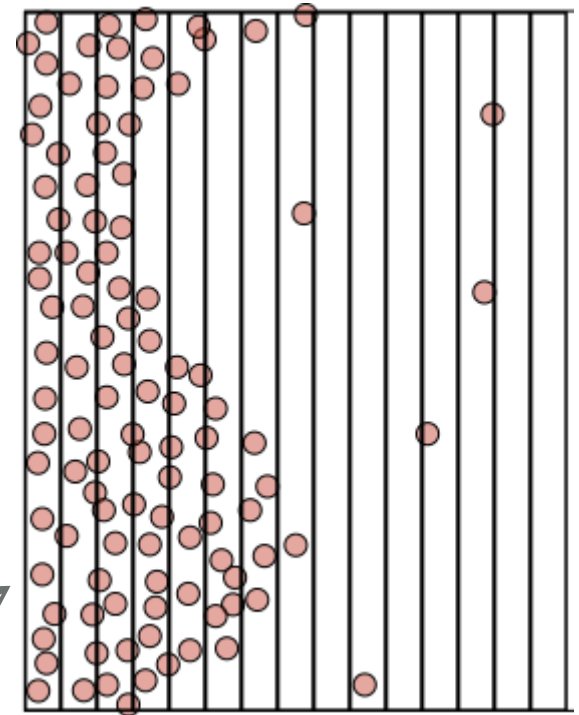
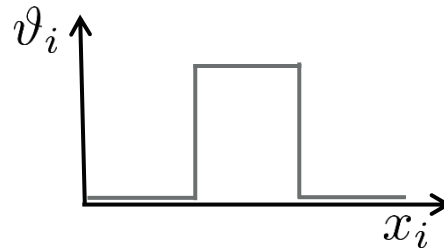


Getting Quantities Near an Interface

- Density in a cubic control volume

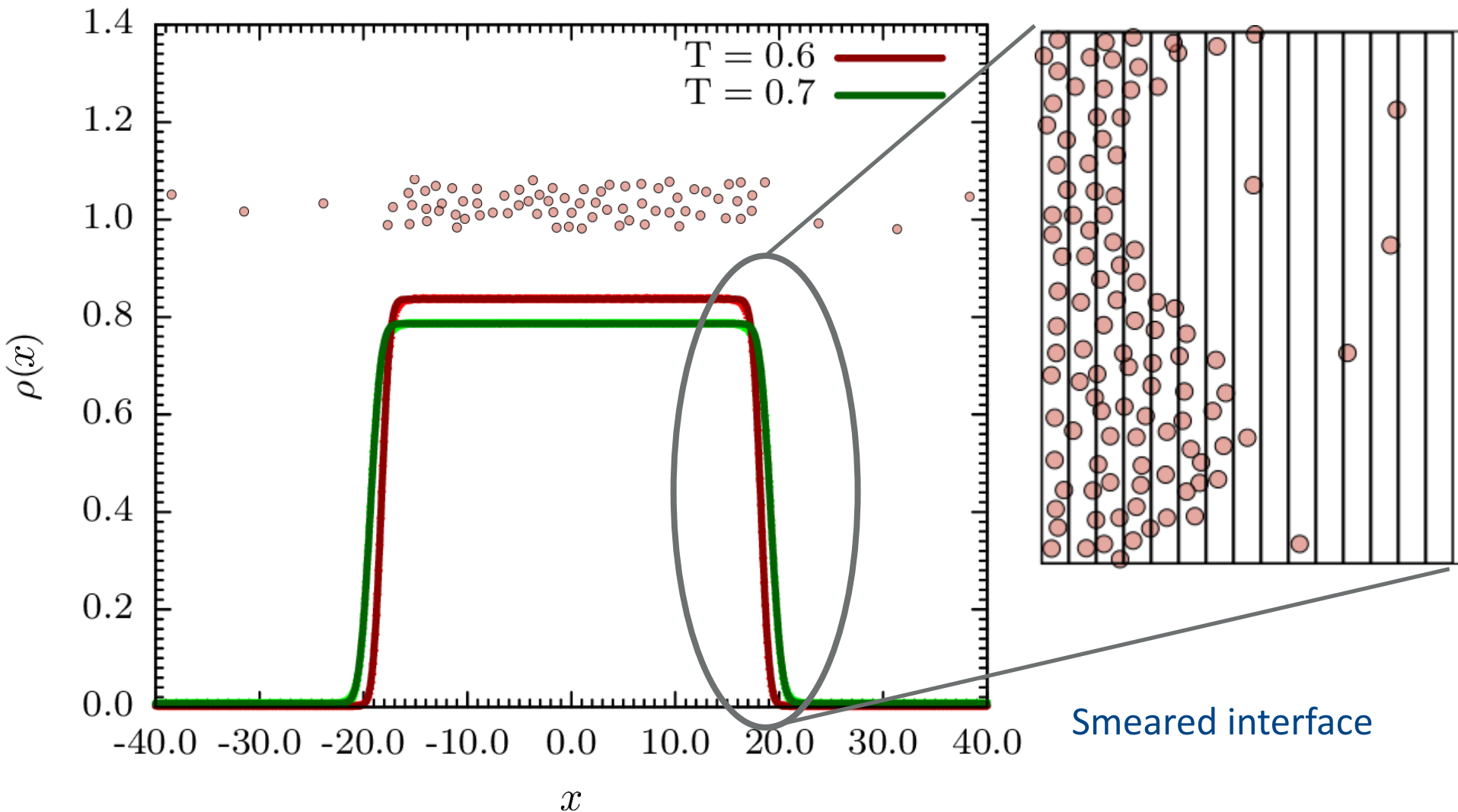
$$\int_V \rho(\mathbf{r}, t) dV = \sum_{i=1}^N m_i \int_V \delta(\mathbf{r} - \mathbf{r}_i) dV$$

$$= \sum_{i=1}^N m_i [H(x^+ - x_i) - H(x^- - x_i)]$$



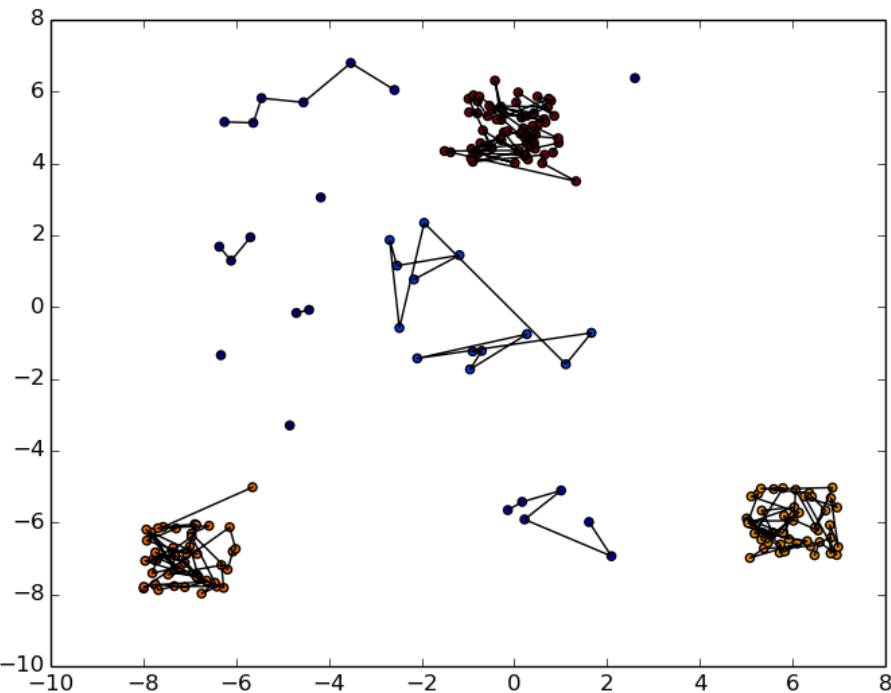
- Top hat function selects molecules inside a volume. Domain is therefore split into uniform bins in x

Results for Density

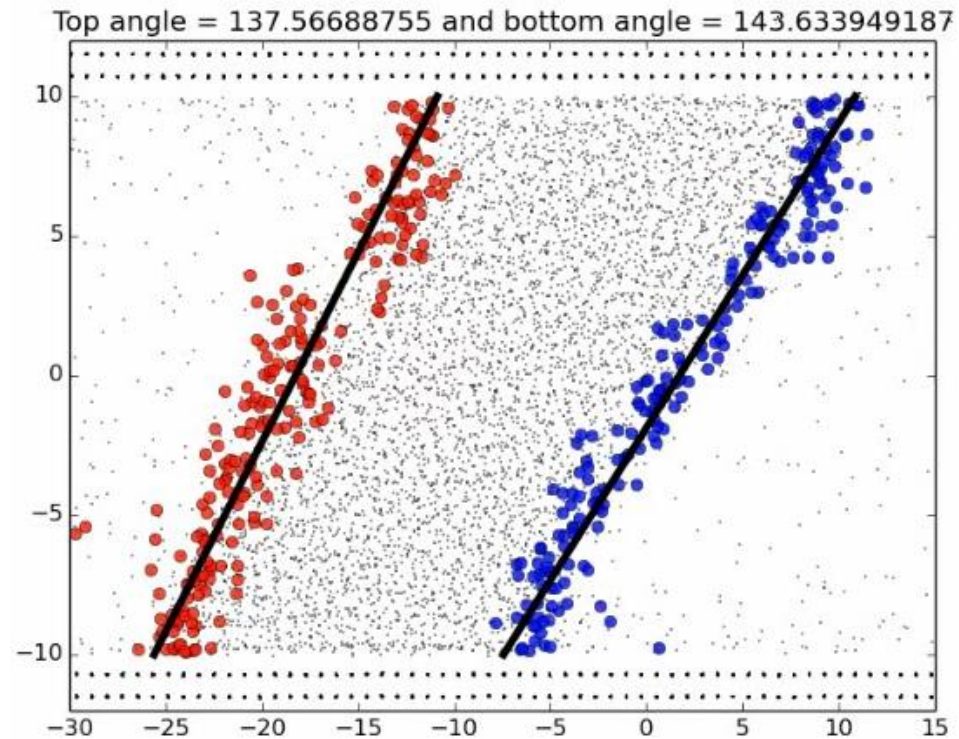


Cluster analysis and surface fitting

Cluster analysis

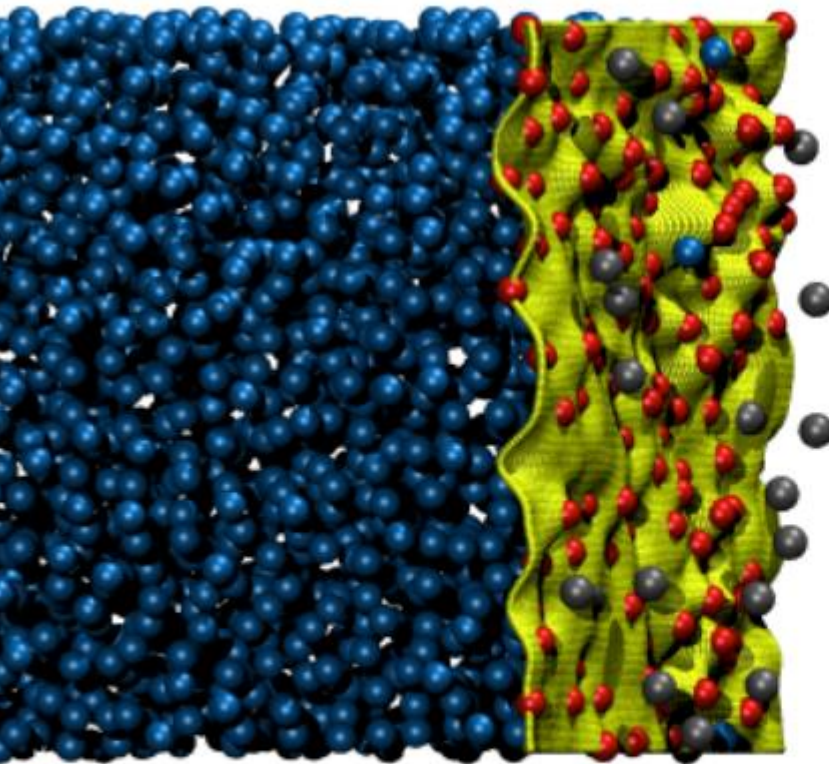


Finding the fluid-liquid interface



Intrinsic surface

- Choose a function which can fit down to intermolecular spacing



$$W = \frac{1}{2} \sum_{i=1}^{N_s} (x_i - \xi(y, z))^2 + \lambda C$$

$$\xi(y, z) = \sum_{\mu, \nu} a_{\mu\nu} f_{\mu}(x) f_{\nu}(y)$$

Function of sines and cosines

An Intrinsic Control Volume Functional

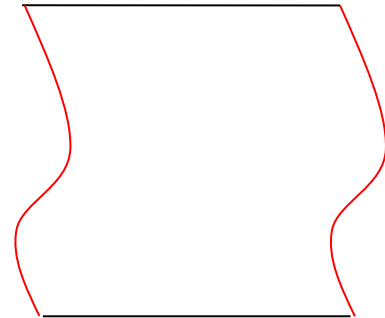
- Integral between moving intrinsic surfaces

$$\vartheta_i \equiv \int_{z^-}^{z^+} \int_{y^-}^{y^+} \int_{x^- + \xi(y,z)}^{x^+ + \xi(y,z)} \delta(x - x_i) \delta(y - y_i) \delta(z - z_i) dx dy dz$$

$$= [H(x^+ + \xi(y_i, z_i) - x_i) - H(x^- + \xi(y_i, z_i) - x_i)]$$

$$\times [H(y^+ - y_i) - H(y^- - y_i)]$$

$$\times [H(z^+ - z_i) - H(z^- - z_i)]$$



- In words

$$\vartheta \equiv \begin{cases} 1 & \text{if molecule is inside volume} \\ 0 & \text{if molecule is outside volume} \end{cases}$$

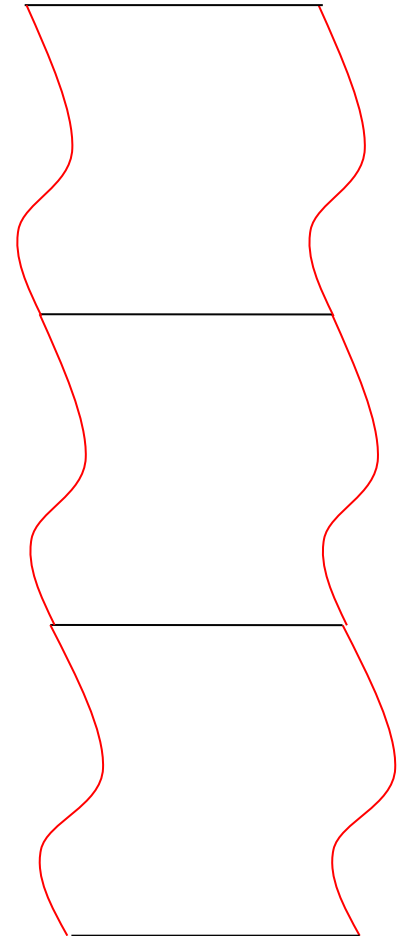
An Intrinsic Control Volume Functional

- Assume a periodic domain in y and z with a surface which is equal at the top and bottom (correct as sines and cosines)

$$\begin{aligned} \vartheta_i = & \left[H(x^+ + \xi(y_i, z_i) - x_i) - H(x^- + \xi(y_i, z_i) - x_i) \right] \\ & \times \left[H(y^+ - y_i) - H(y^- - y_i) \right] \\ & \times \left[H(z^+ - z_i) - H(z^- - z_i) \right] \end{aligned}$$

- In words

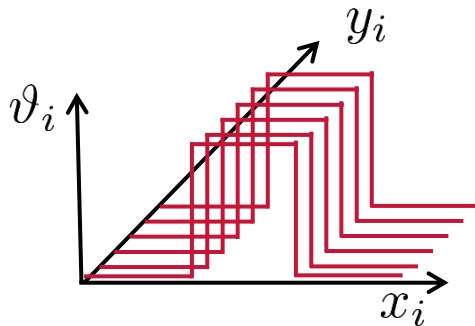
$$\vartheta \equiv \begin{cases} 1 & \text{if molecule is inside volume} \\ 0 & \text{if molecule is outside volume} \end{cases}$$



An Intrinsic Control Volume Functional

- Assume a periodic domain in y and z with a surface which is equal at the top and bottom (correct as sines and cosines)

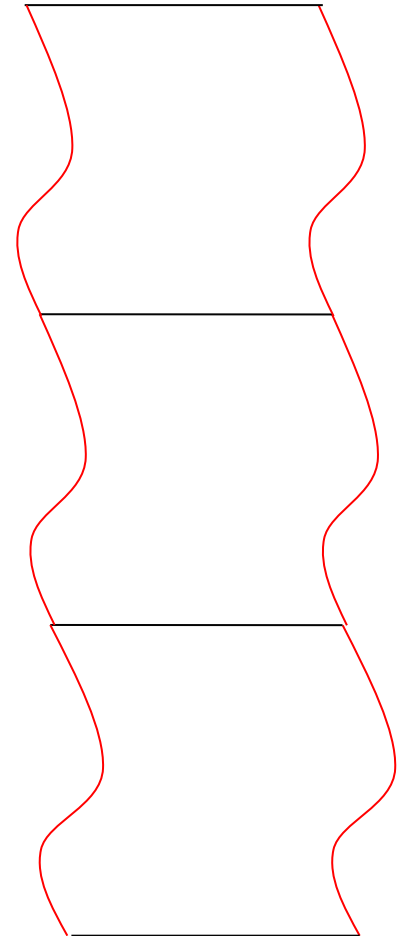
$$\vartheta_i = [H(x^+ + \xi(y_i, z_i) - x_i) - H(x^- + \xi(y_i, z_i) - x_i)]$$



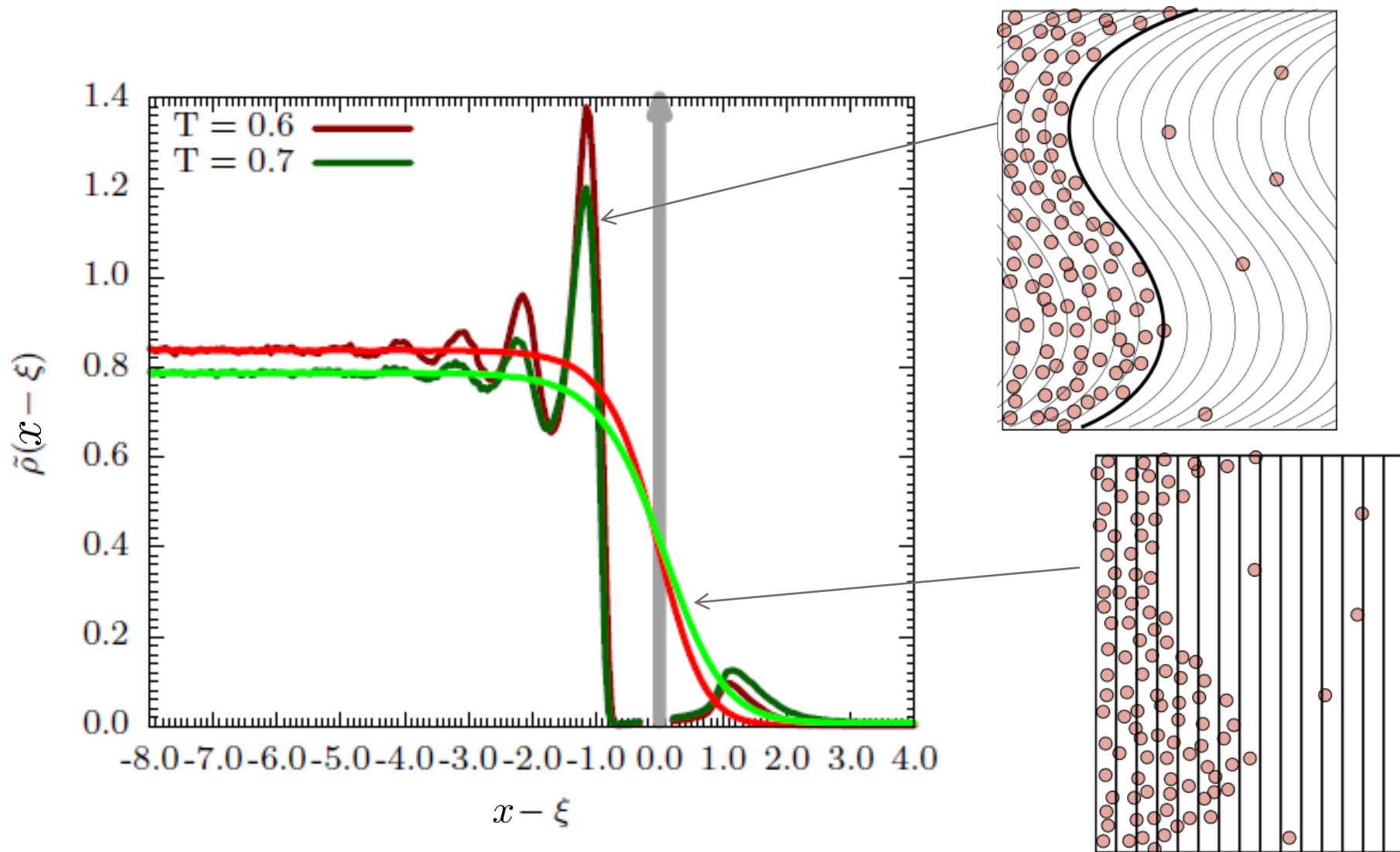
- Top hat functions (varying with y) select molecules inside a volume.

- In words

$$\vartheta \equiv \begin{cases} 1 & \text{if molecule is inside volume} \\ 0 & \text{if molecule is outside volume} \end{cases}$$



Results for Density



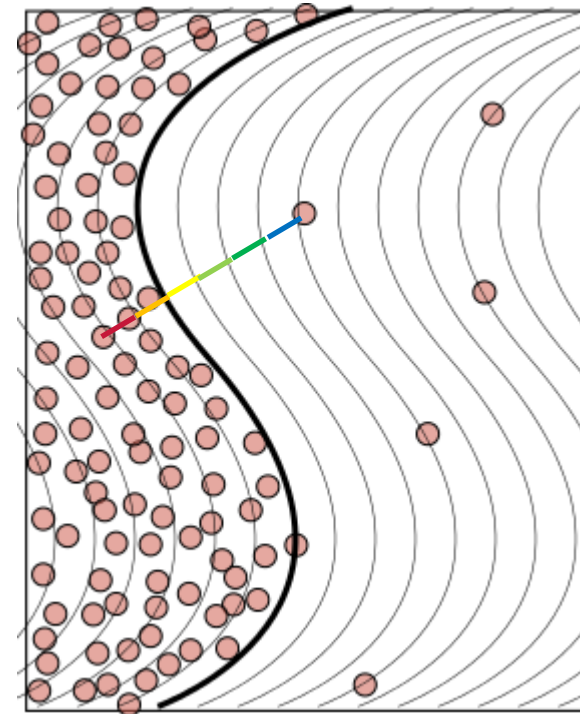
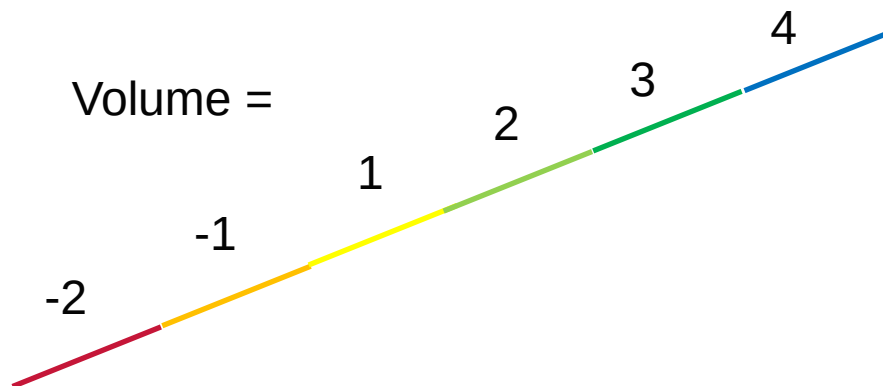
Applied to Stress

- Stress in a control volume based on the intrinsic surface

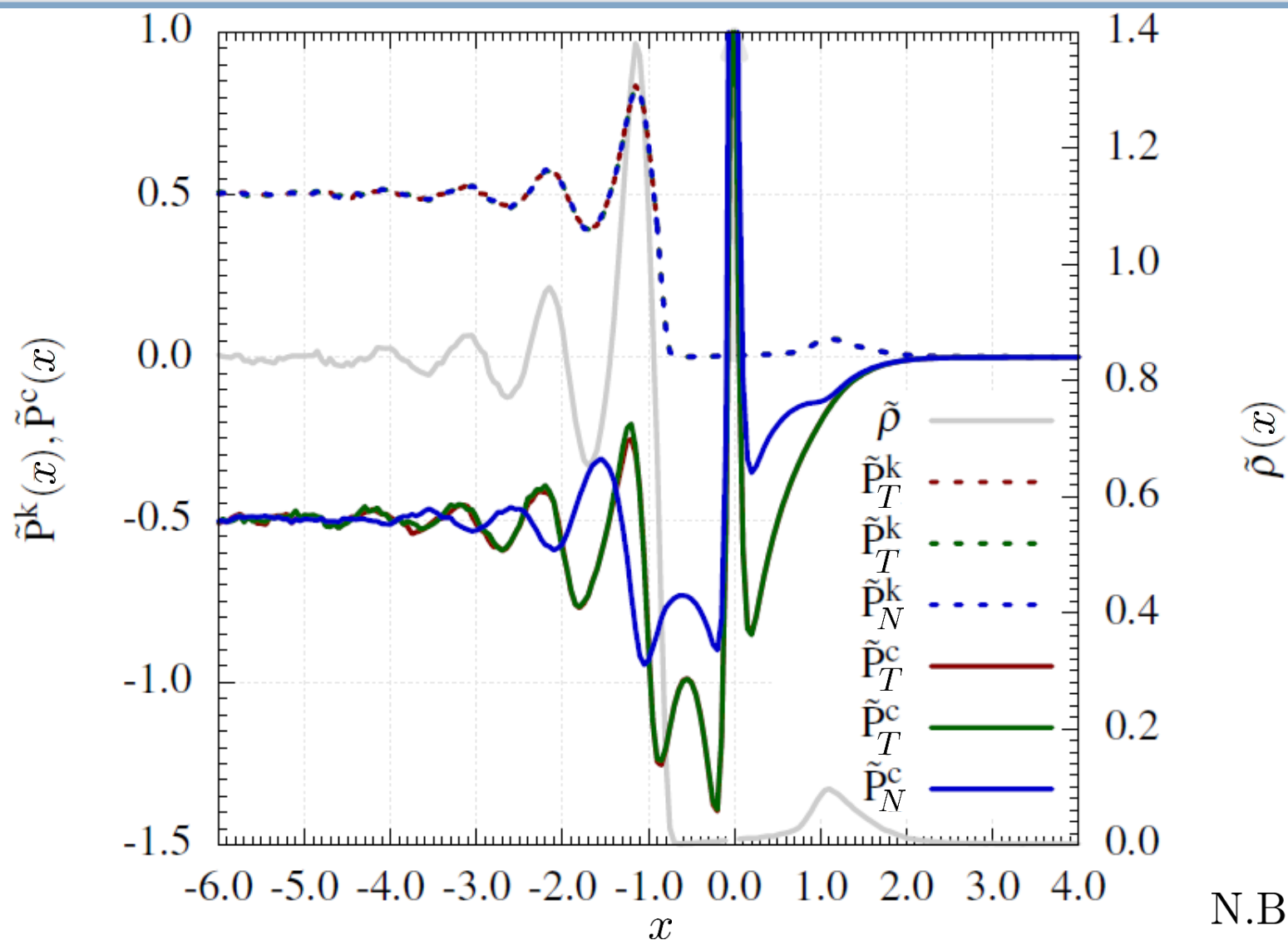
$$\int_V \tilde{\Pi}^c dV = \sum_{i,j}^N \mathbf{f}_{ij} \mathbf{r}_{ij} \int_0^1 \vartheta_s ds$$

$$\vartheta_s = \begin{bmatrix} H(x^+ + \xi(y_s, z_s) - x_s) \\ -H(x^- + \xi(y_s, z_s) - x_s) \end{bmatrix}$$

- Line of interaction split over every volume it passes through (shown here by colour)



Results for Stress

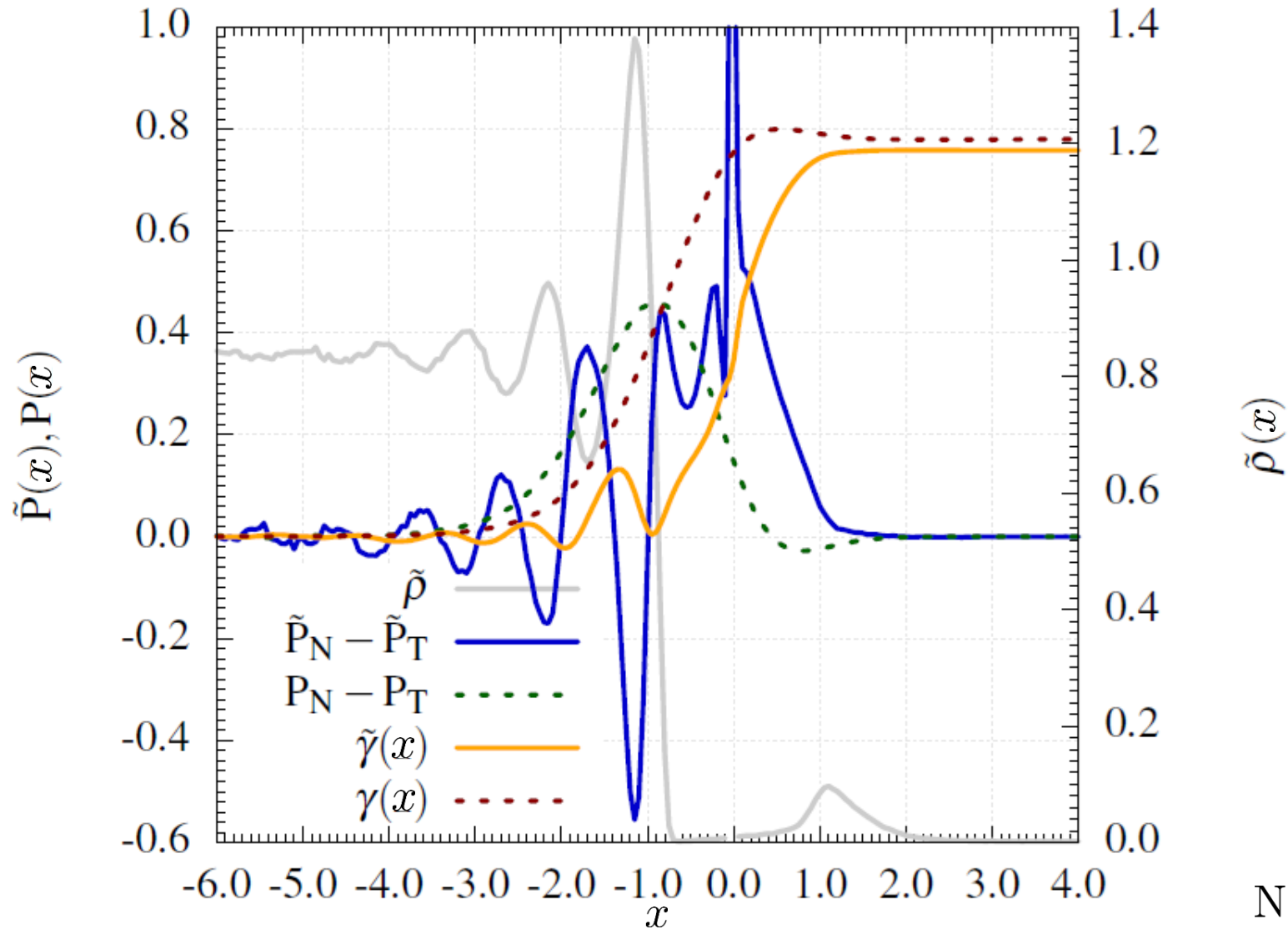


N.B. $\Pi \equiv P$

$$\gamma = \int_{-\infty}^x [\Pi_N - \Pi_T] dx$$

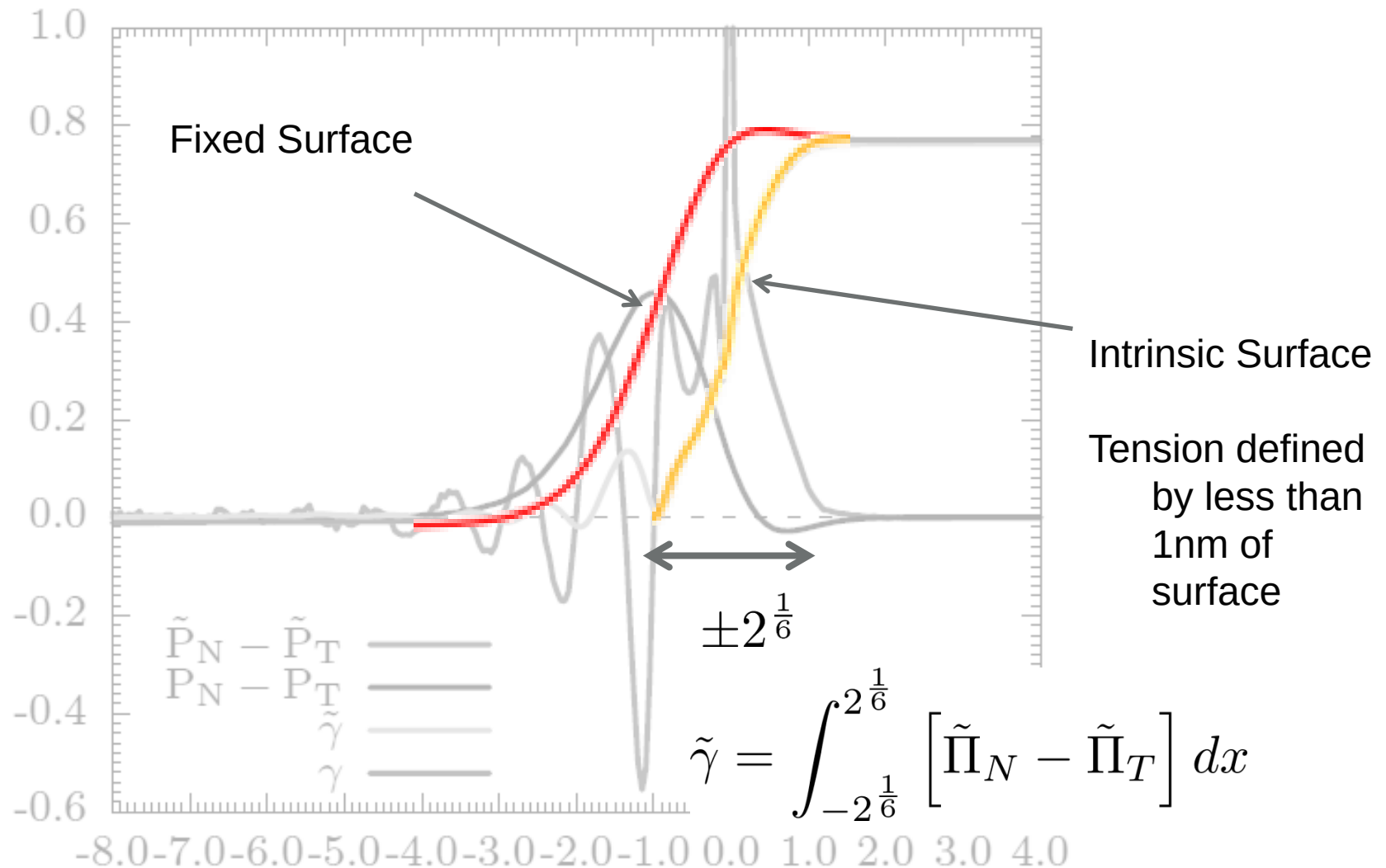


Results for Surface Tension



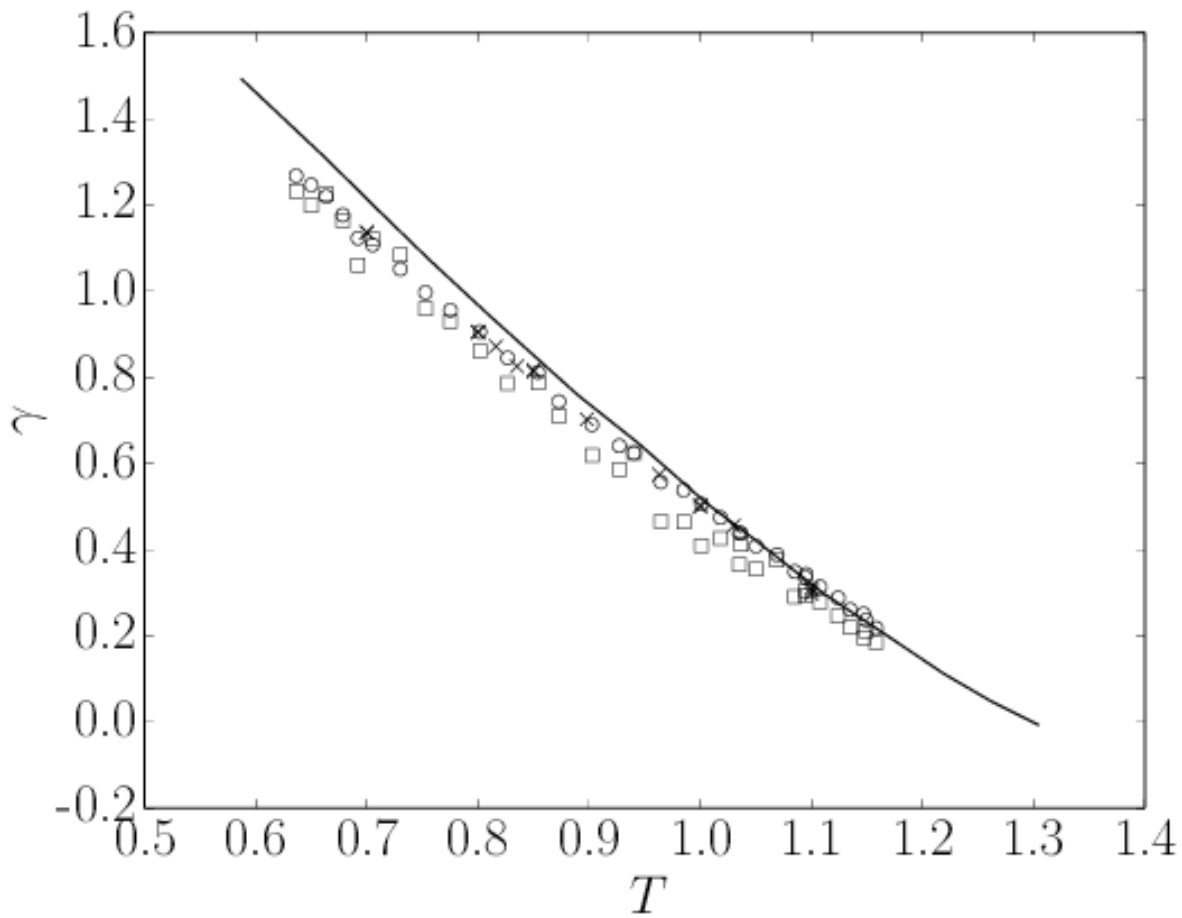
N.B. $\Pi \equiv P$

Results for Surface Tension



Validation Against Experiments

- Surface tension matched to results for liquid Argon using both intrinsic surface and flat surface



- Kirkwood Buff to get surface tension from normal and tangential stresses

$$\gamma = \int_{-\infty}^x [\Pi_N - \Pi_T] dx$$

Overview

Control volume for intrinsic surface

$$\vartheta = \int_{V(t)} \delta(\mathbf{r} - \mathbf{r}_i) dV$$

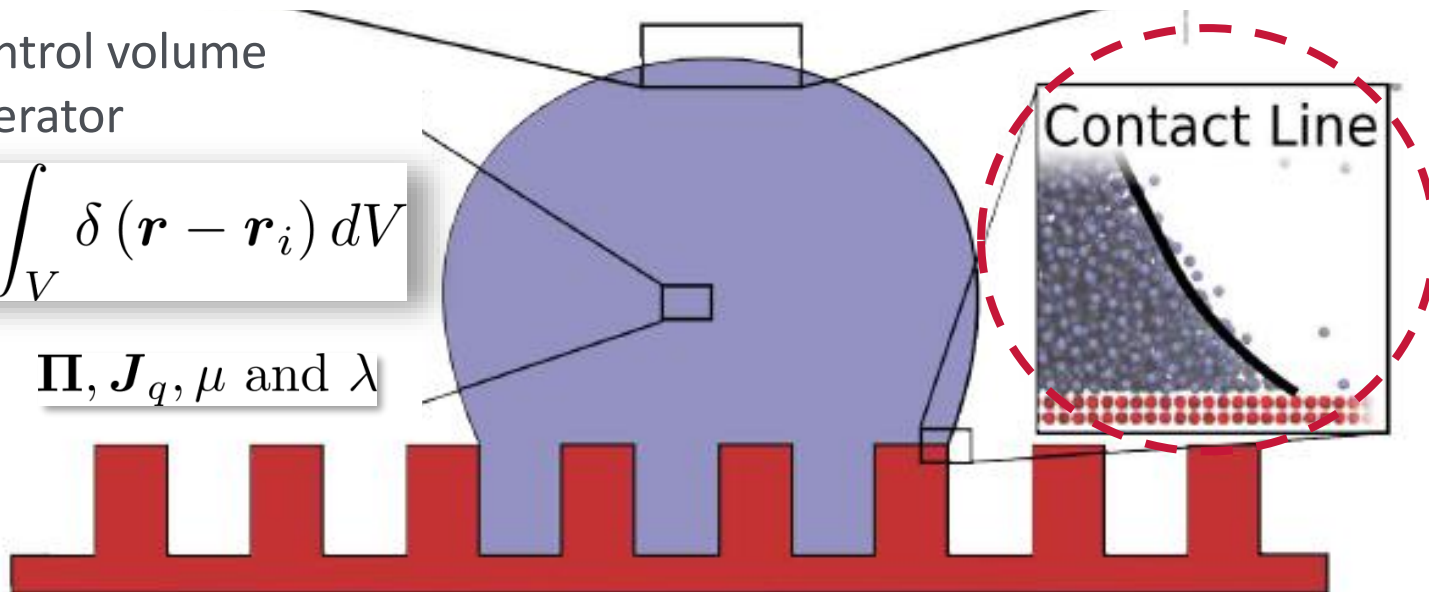
gets $\tilde{\Pi}, \tilde{J}_q$ and γ

Volume moving with fitted
spectral surface to
molecular spacing

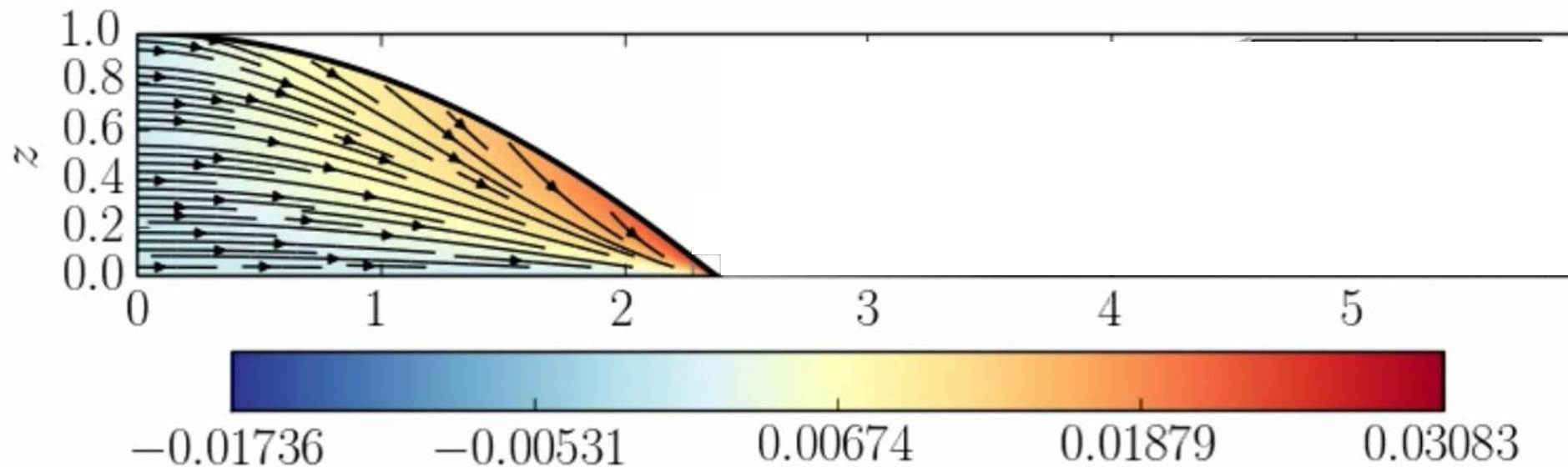
Control volume
operator

$$\vartheta = \int_V \delta(\mathbf{r} - \mathbf{r}_i) dV$$

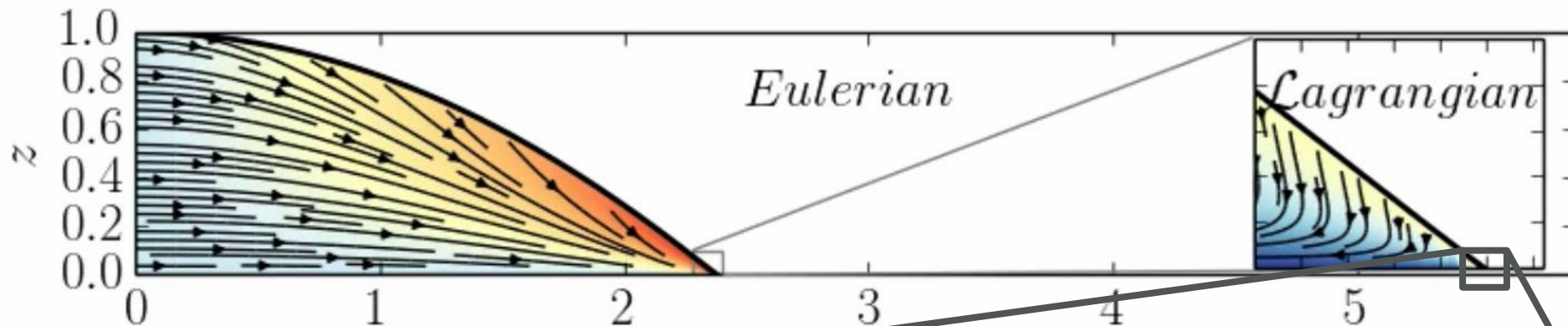
gets Π, J_q, μ and λ



Coupled Droplet Spreading and MD

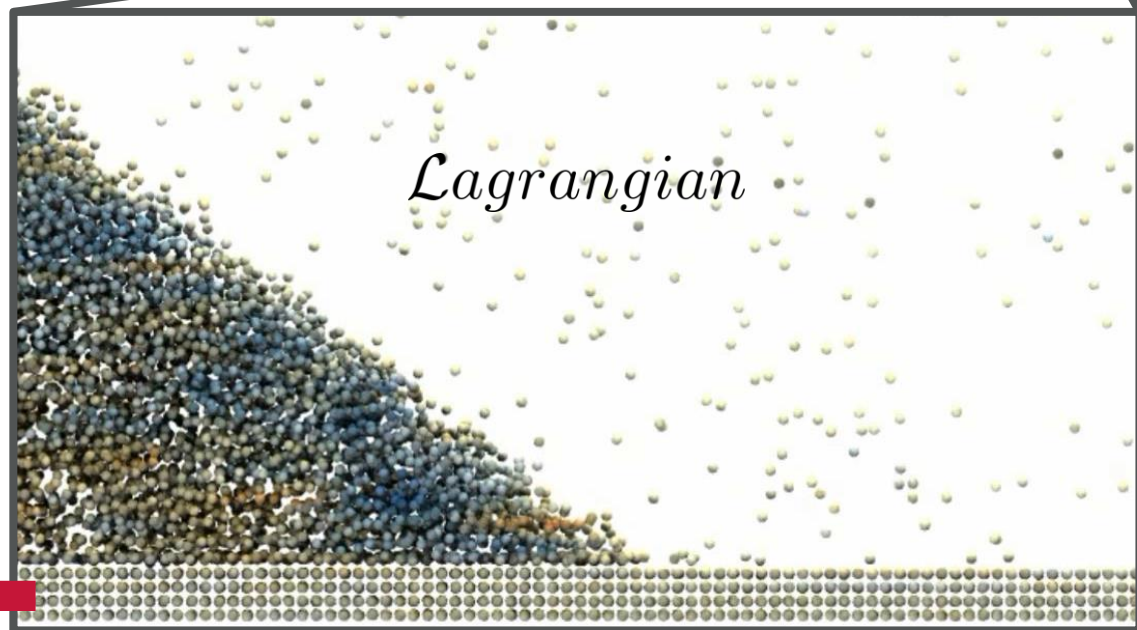


Coupled Droplet Spreading and MD



- Model the moving contact line with MD
- We want contact line speed as a function of continuum contact angle

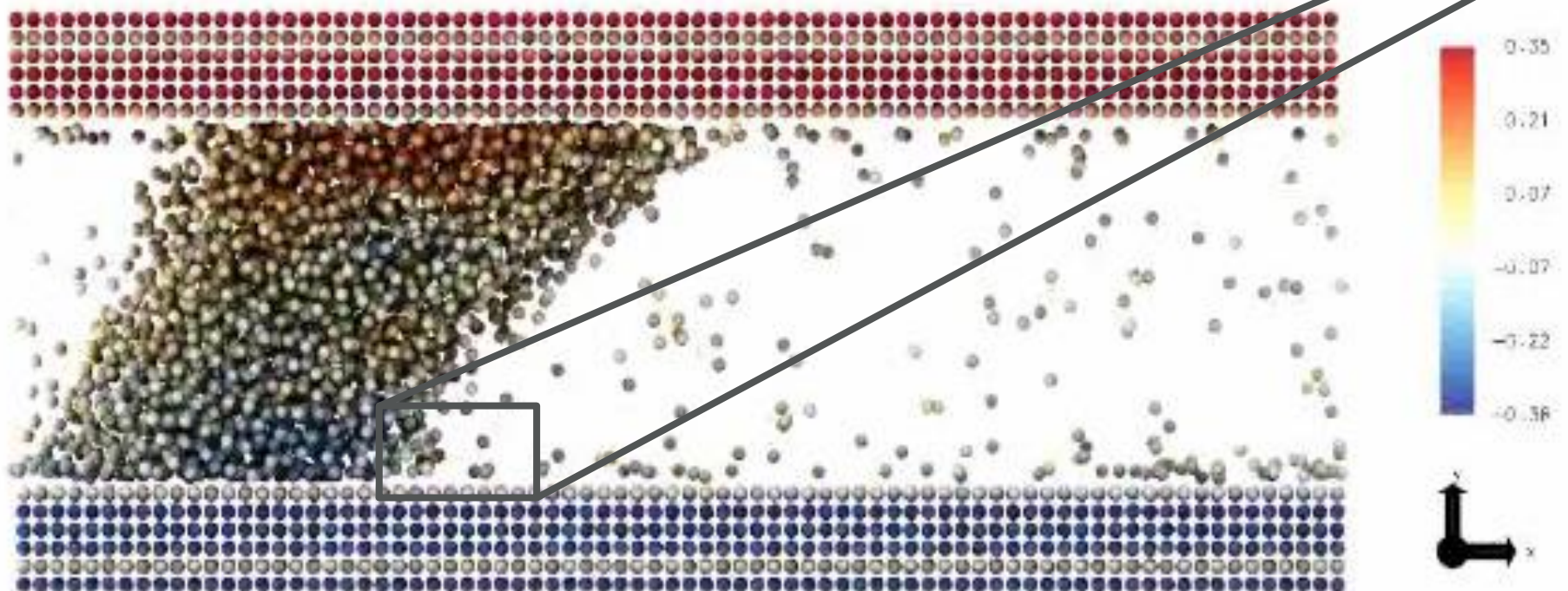
$$\frac{dx_c}{dt}$$





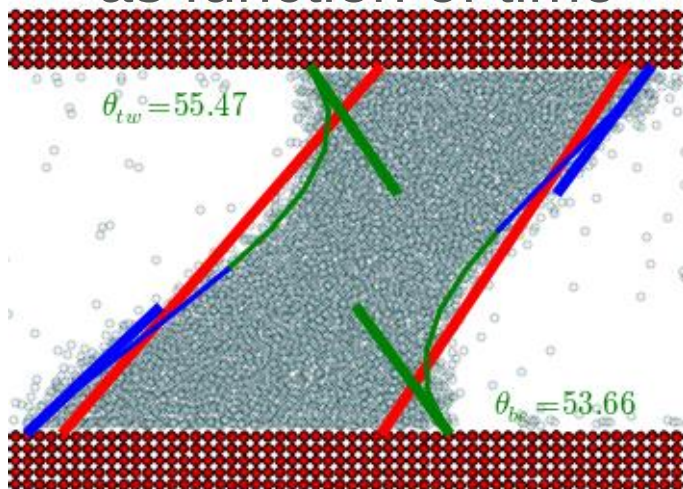
Sheared Liquid Bridge

- Two fluid phases and sliding molecular walls
- Simple test case to explore wall velocity vs contact line angle
- Non-Equilibrium Steady State

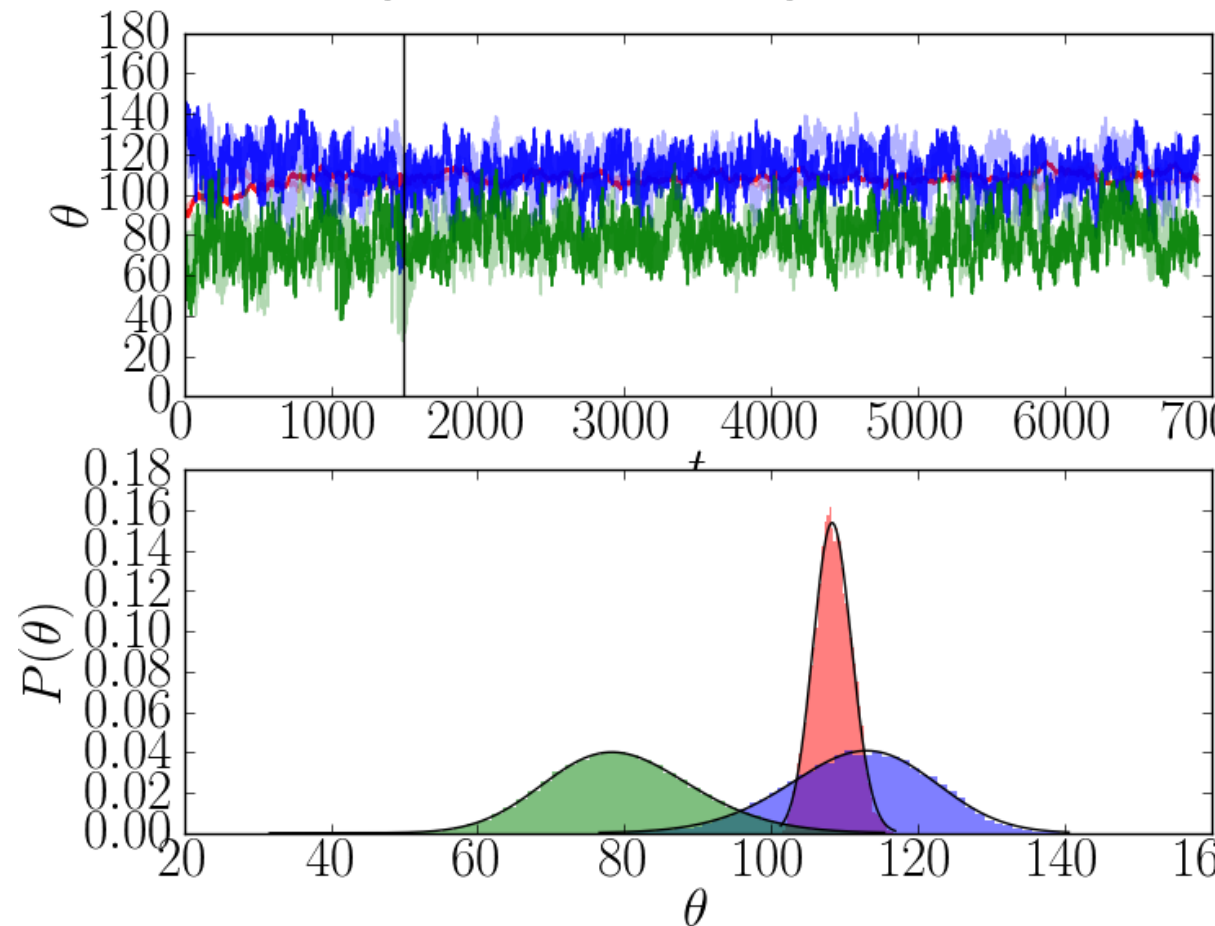


Time Evolution of Contact Angle

- Fit (simple) surface to cluster and get contact angle evolution as function of time



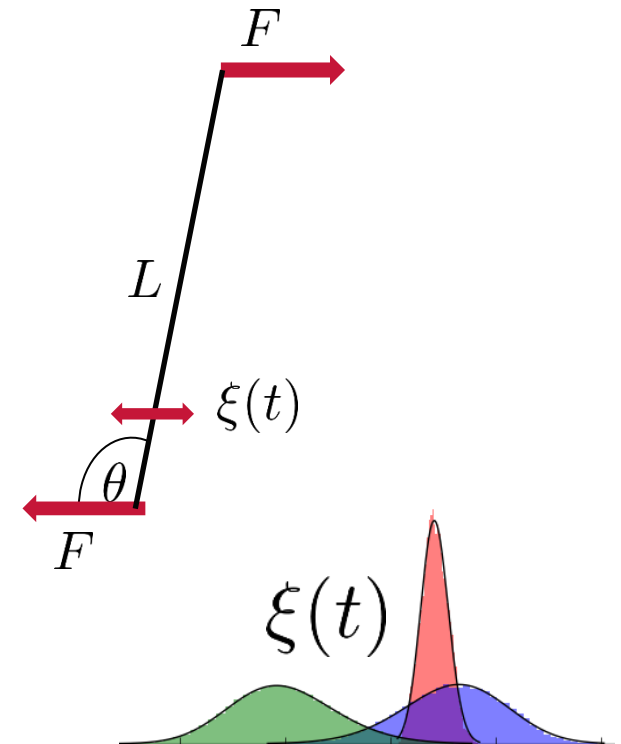
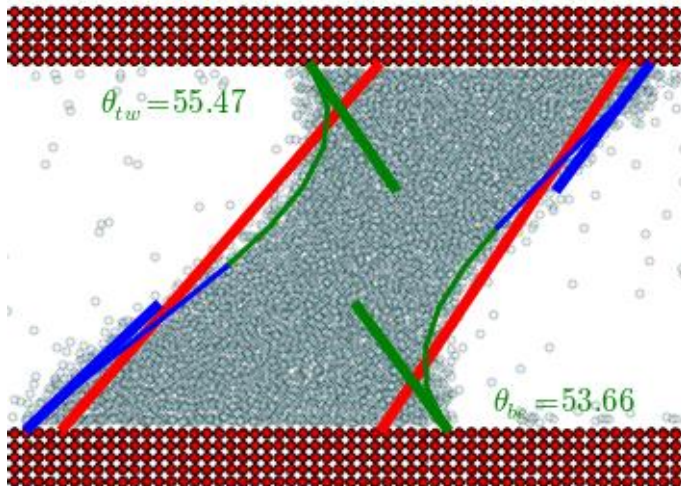
- Probability density function of angle shows range of micro-scale behaviour



- Linear**, **Advancing** and **Receding** angles

Building this into the Continuum Model

- Model the movement of the contact line as a torsional string mass system + a random noise term



$$I\ddot{\theta} + \Gamma\dot{\theta} + k\theta - \xi = \mathcal{T}$$

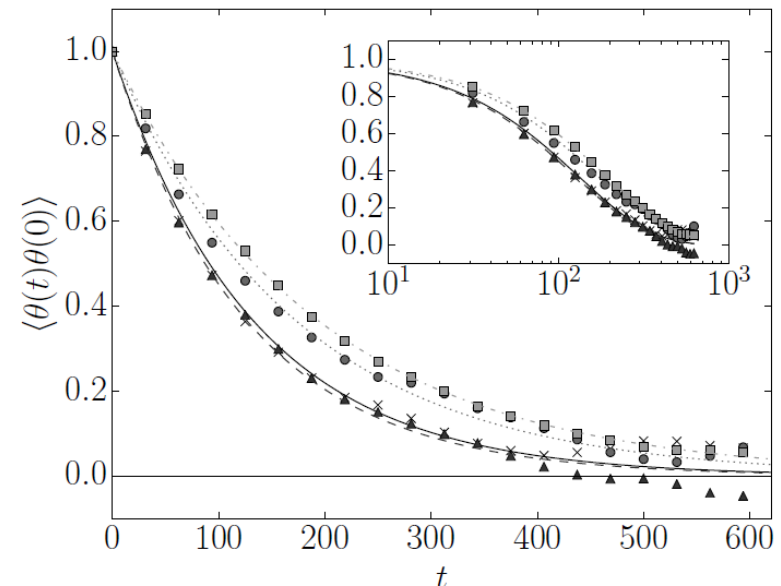
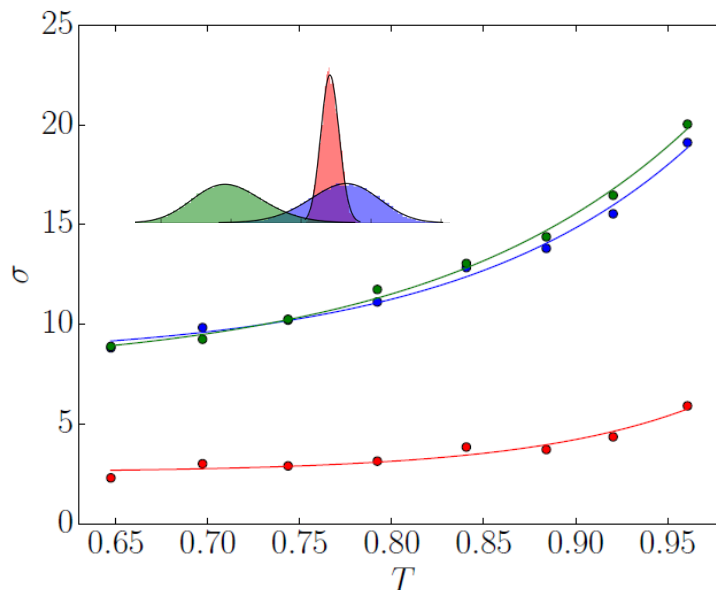
- Torque $\mathcal{T} = F \times L$ approximately equal to wall sliding

Building this into the Continuum Model

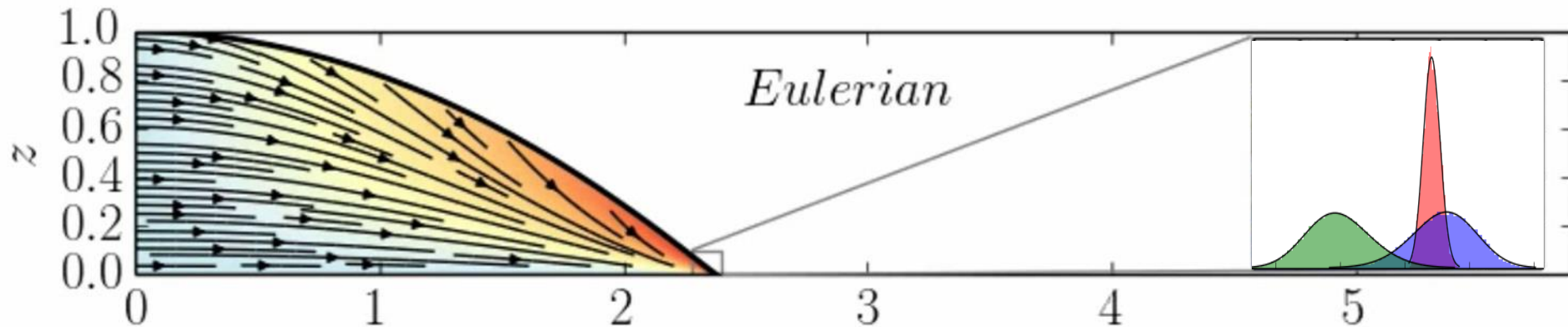
- In the limit overdamped limit we get the Langevin Equation

$$\dot{\theta} + \frac{k}{\Gamma} [\theta - \langle \theta \rangle] - \frac{1}{\Gamma} \xi(t) = 0 \text{ where } \langle \xi(t) \xi(t') \rangle = C \delta(t - t'),$$

- Coefficients parameterised using
 - Standard deviation - function of temperature velocity independent
 - Autocorrelation - roughly velocity and temperature independent.



Building this into the Continuum Model



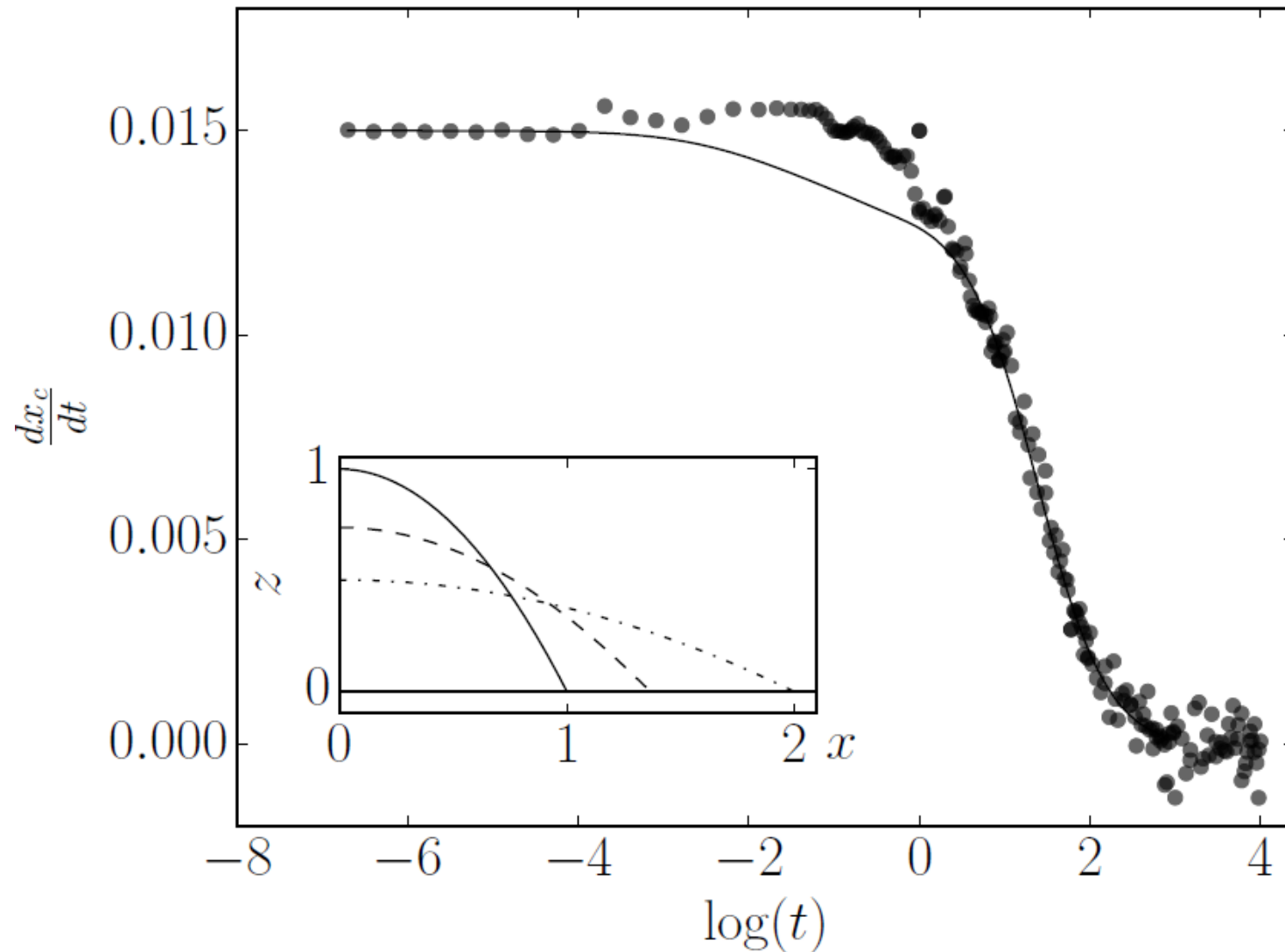
- Speed of contact line obtained from a Langevin model tuned by the molecular PDF

$$\theta^{t+1} = \theta^t - \frac{k\Delta t}{\Gamma} [\theta^t - \langle \theta \rangle] + \xi \frac{\sqrt{C\Delta t}}{\Gamma}$$

*Evolution of the mean
governed by
macroscopic laws
(Tanners, MKT, Cox)*

*Random term
tuned to match
MD contact line
fluctuation and
autocorrelation*

Coupled with CFD model



Overview

Control volume for intrinsic surface

$$\vartheta = \int_{V(t)} \delta(\mathbf{r} - \mathbf{r}_i) dV$$

gets $\tilde{\Pi}, \tilde{J}_q$ and γ

Volume moving with fitted spectral surface to molecular spacing

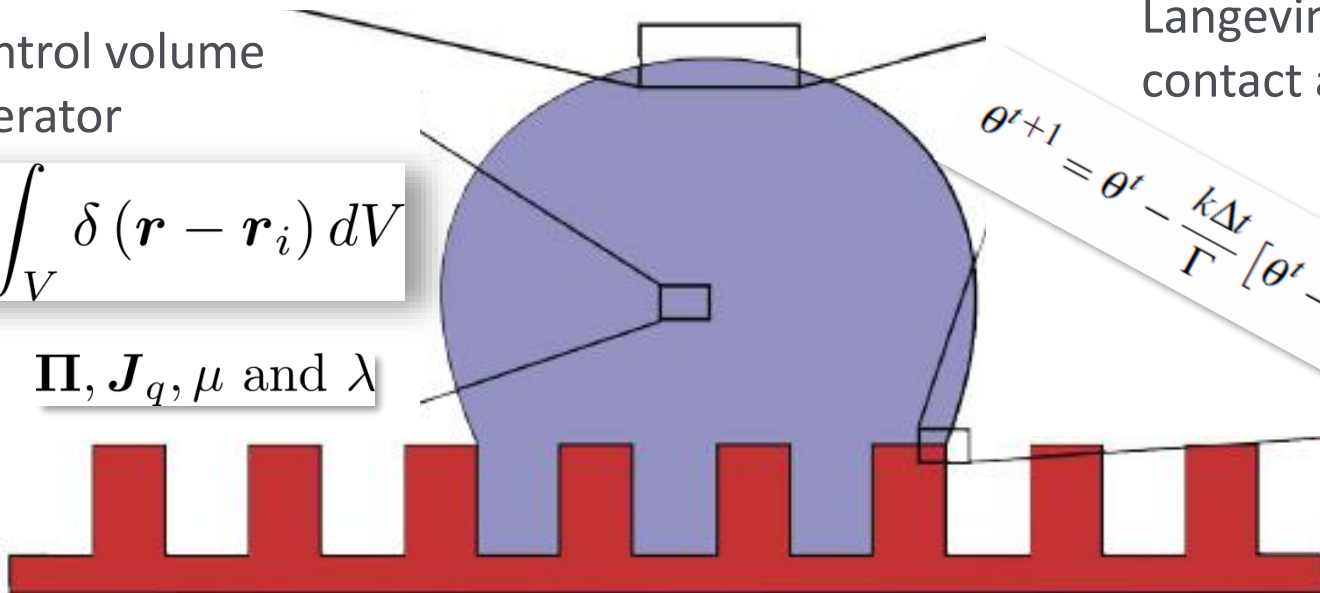
Control volume operator

$$\vartheta = \int_V \delta(\mathbf{r} - \mathbf{r}_i) dV$$

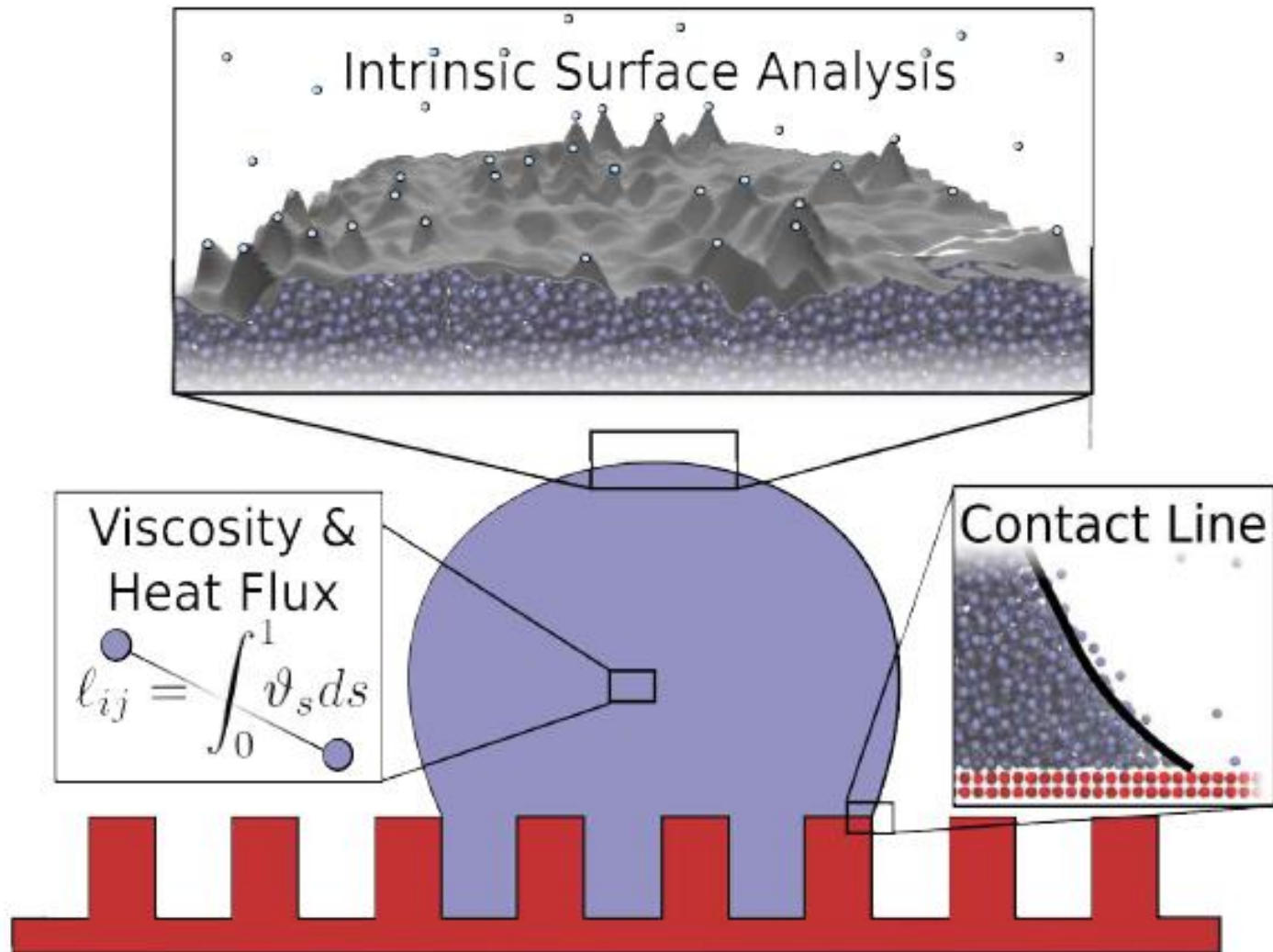
gets Π, J_q, μ and λ

Langevin equation for contact angle

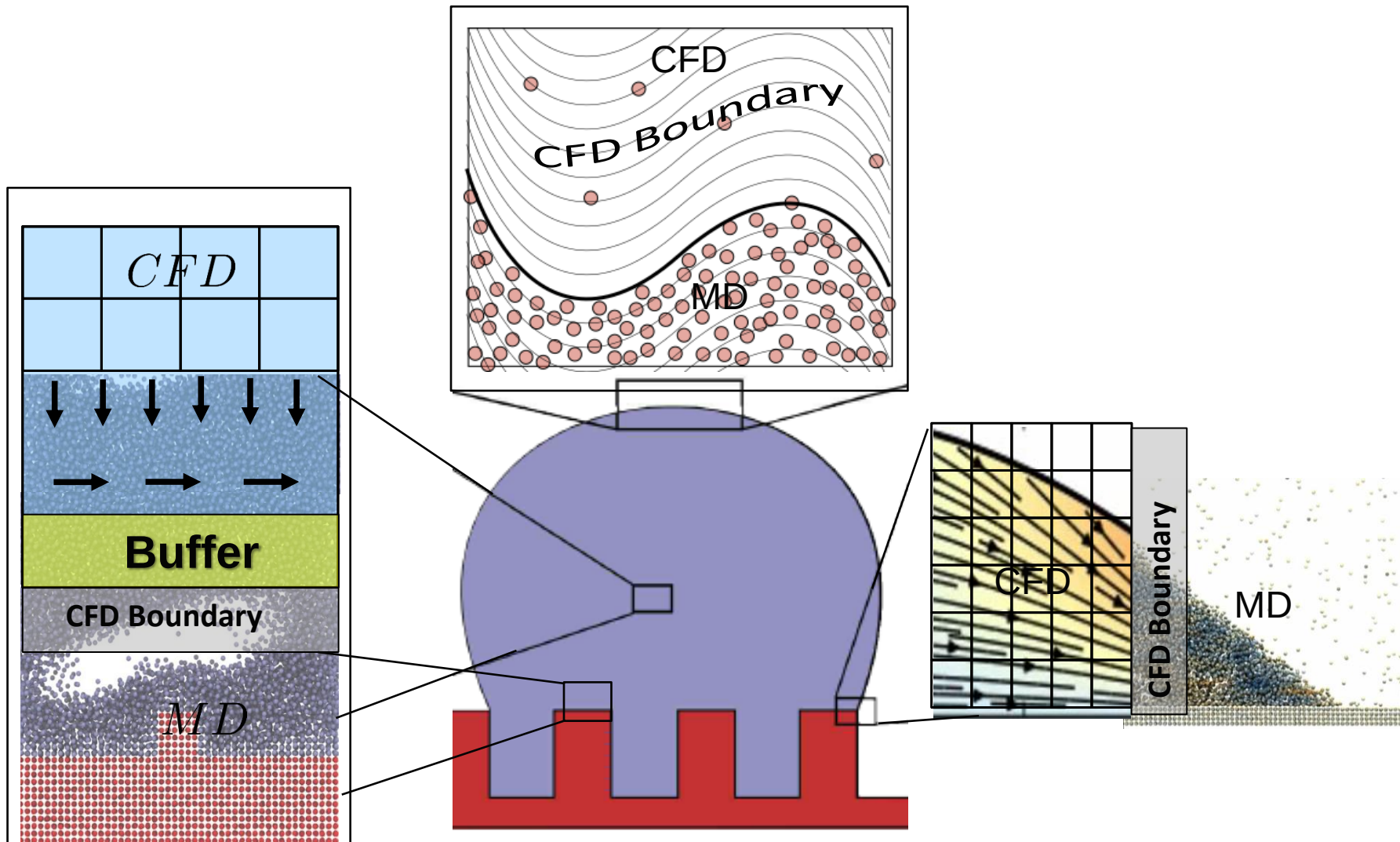
$$\theta^{t+1} = \theta^t - \frac{k\Delta t}{\Gamma} [\theta^t - \langle \theta \rangle] + \xi \frac{\sqrt{C\Delta t}}{\Gamma}$$



Overview



Longer Term Plans - Direct Coupling



Coupled CFD-MD Simulation

- Control Volume Solver

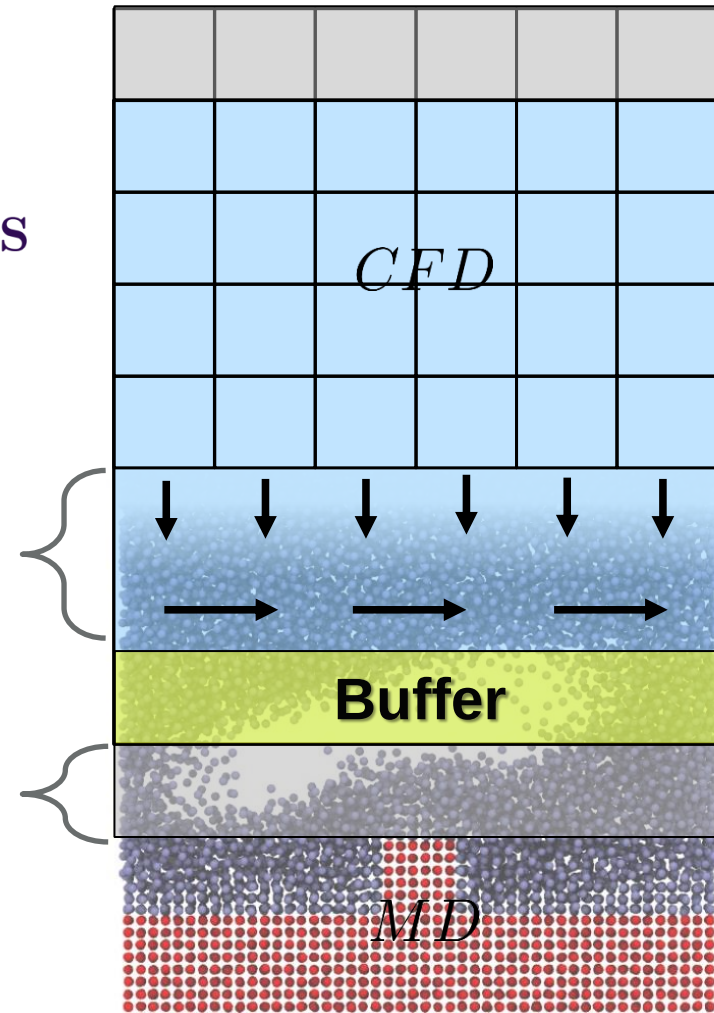
$$\frac{\partial}{\partial t} \int_V \rho \mathbf{u} dV = - \oint_S \rho \mathbf{u} \mathbf{u} \cdot d\mathbf{S} - \oint_S \mathbf{\Pi} \cdot d\mathbf{S}$$

$$m_i \ddot{\mathbf{r}}_i = \mathbf{F}_i + \mathbf{F}_i^C \quad i \in \text{cell} \quad \text{CFD} \rightarrow \text{MD} \text{ Boundary condition}$$

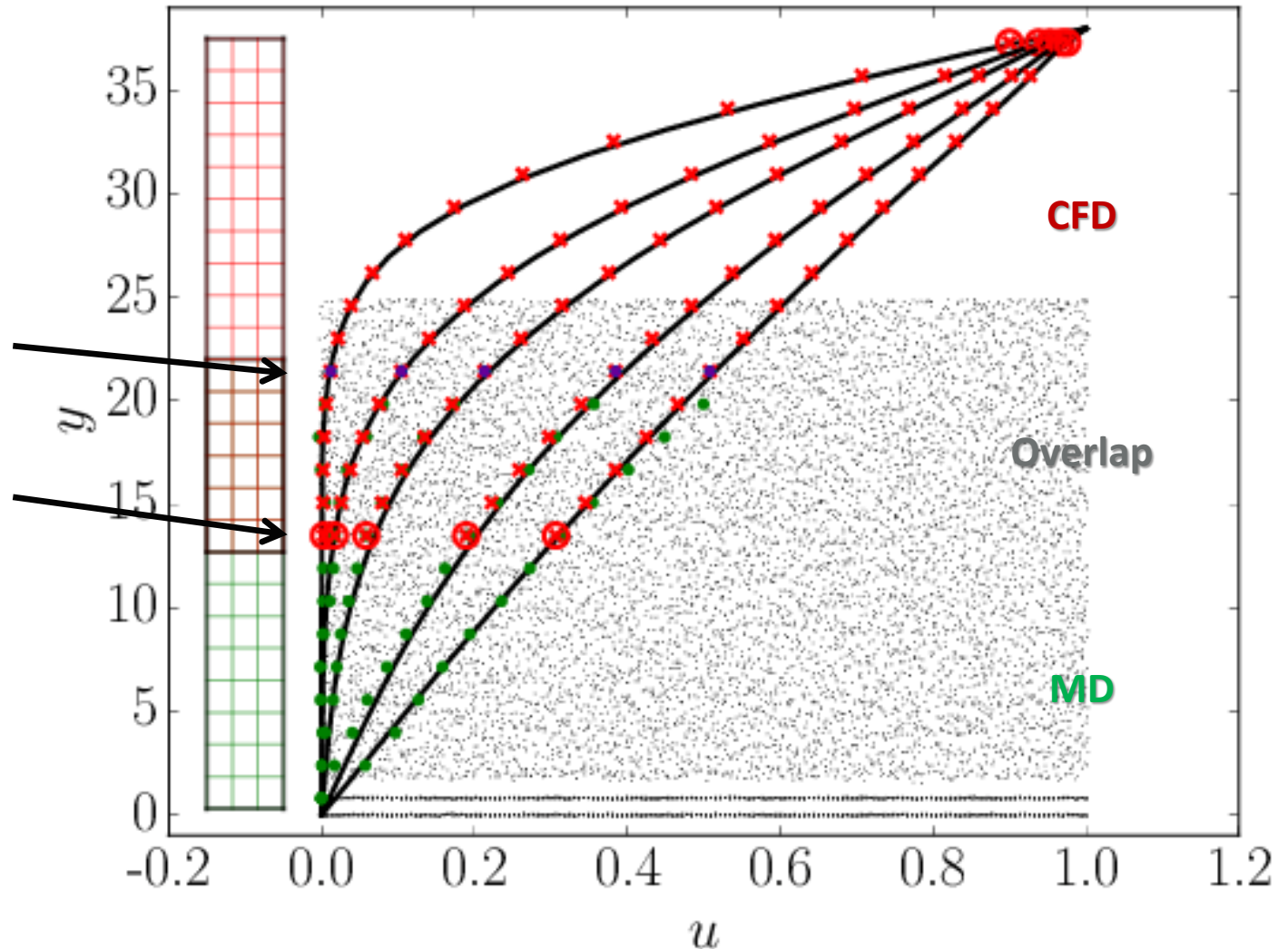
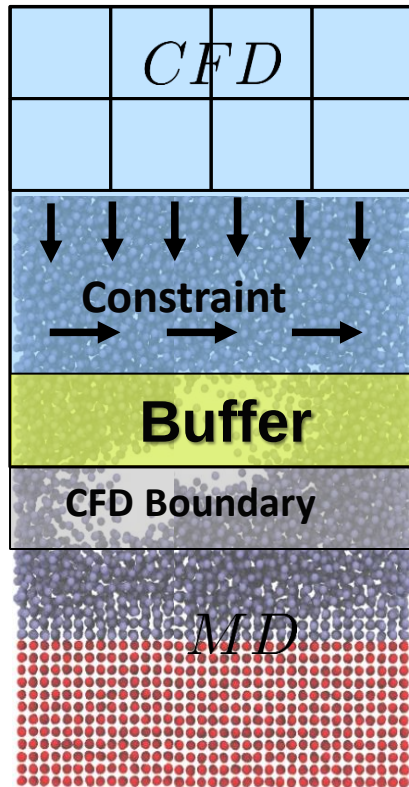
$$\mathbf{u}^{BC} = \sum_{i \in \text{cell}} \mathbf{v}_i \quad \text{MD} \rightarrow \text{CFD} \text{ Boundary condition}$$

- Discrete molecules

$$m_i \ddot{\mathbf{r}}_i = \mathbf{F}_i \text{ for all } i \text{ in } N$$



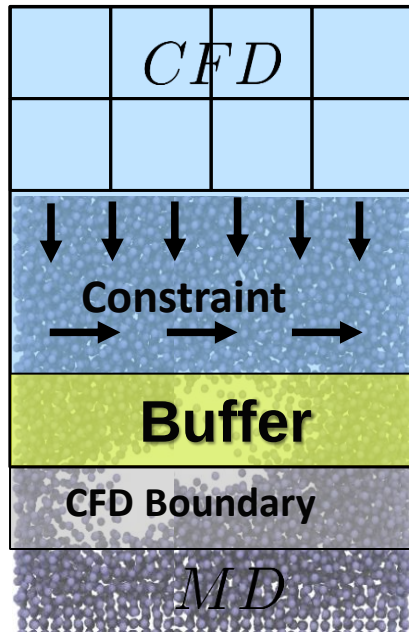
Coupling Results – Couette Flow



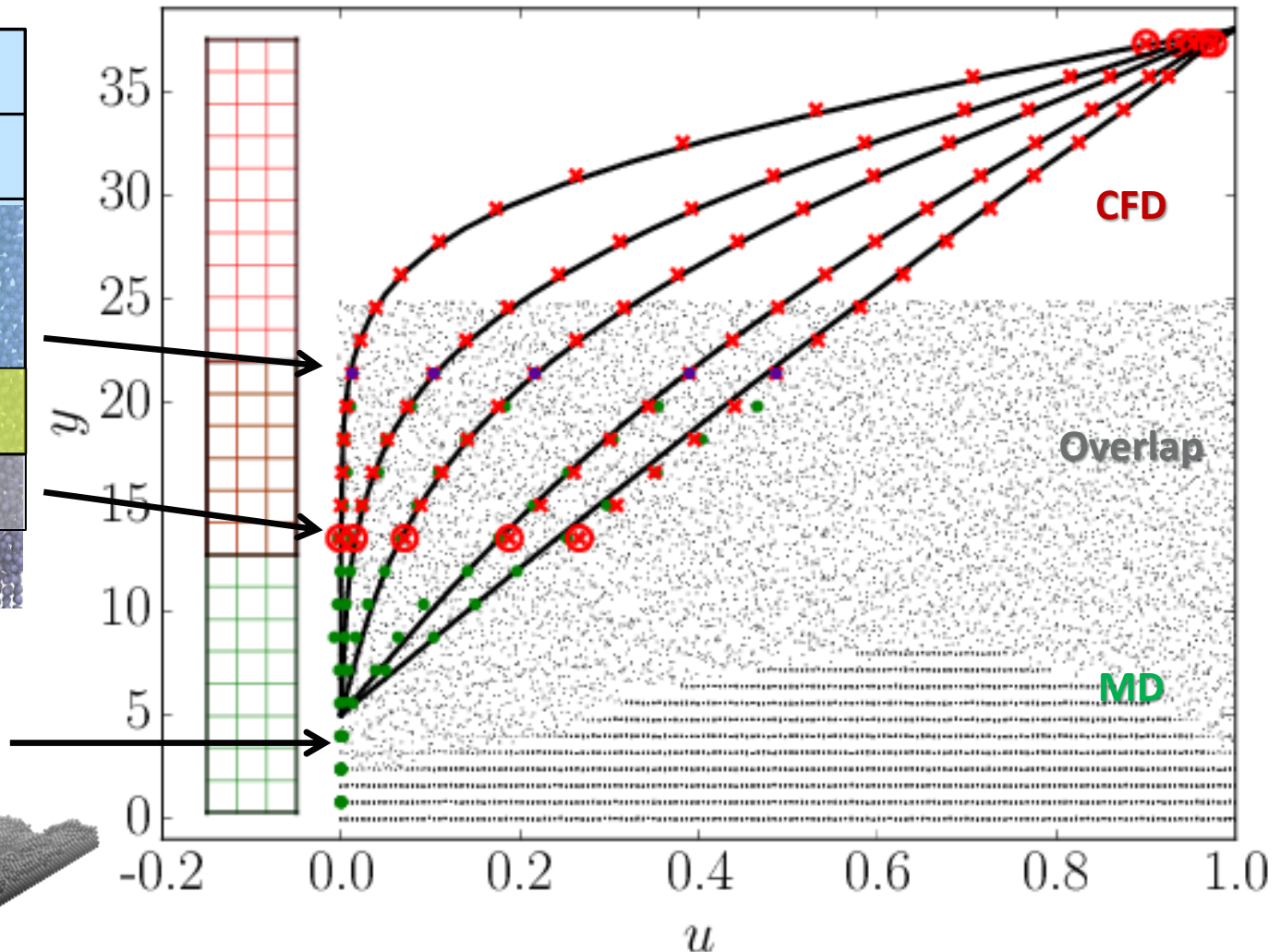
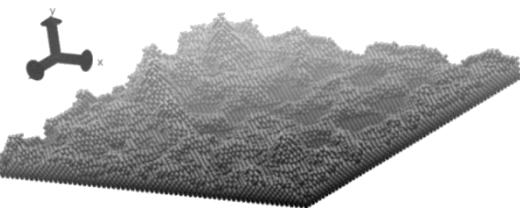
Coupling Results – Couette Flow



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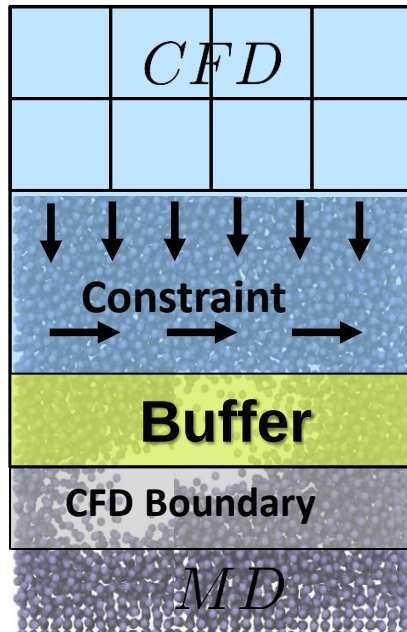


Rough wall shifts
zero location

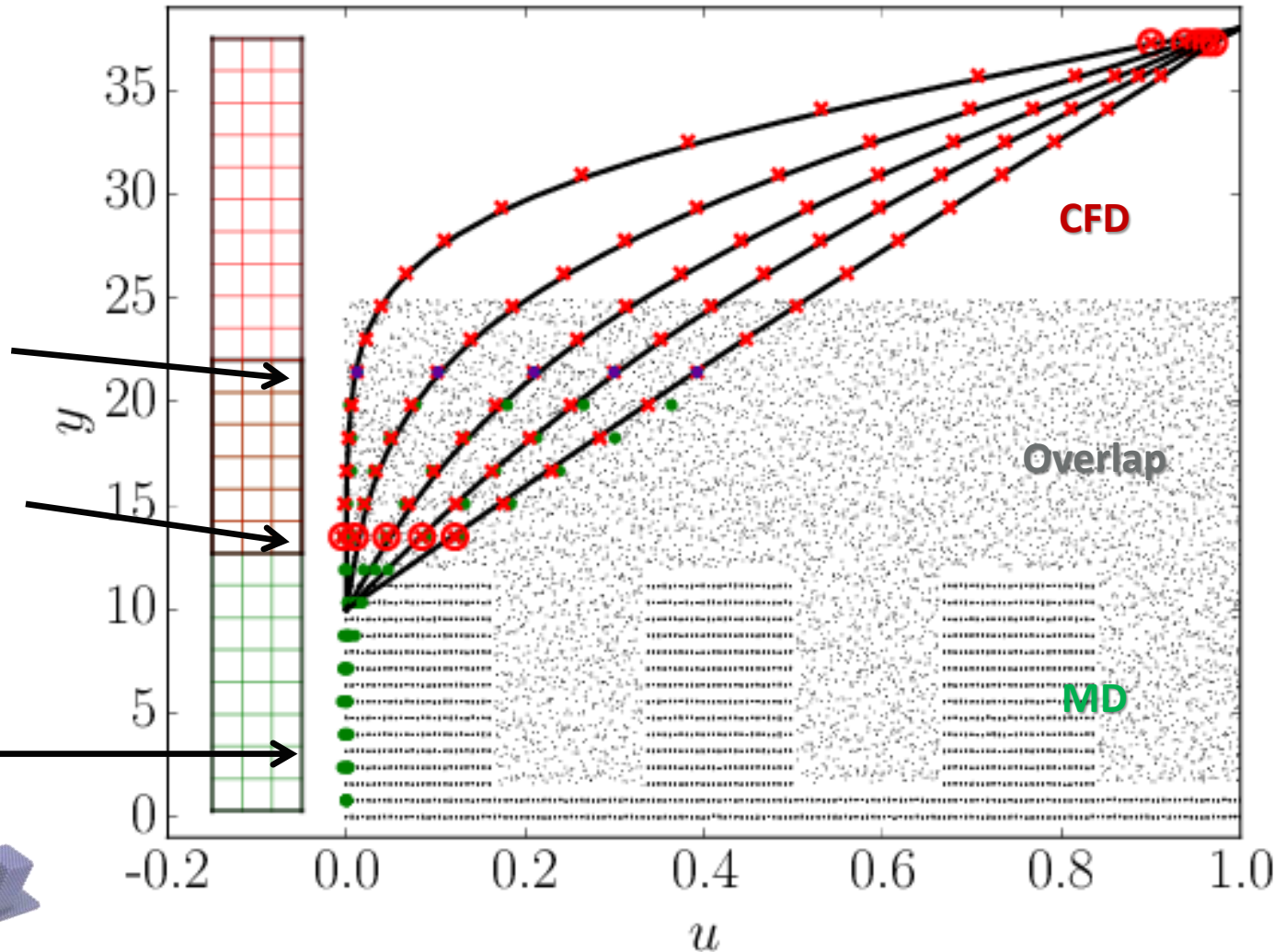
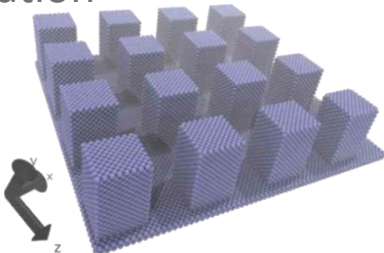


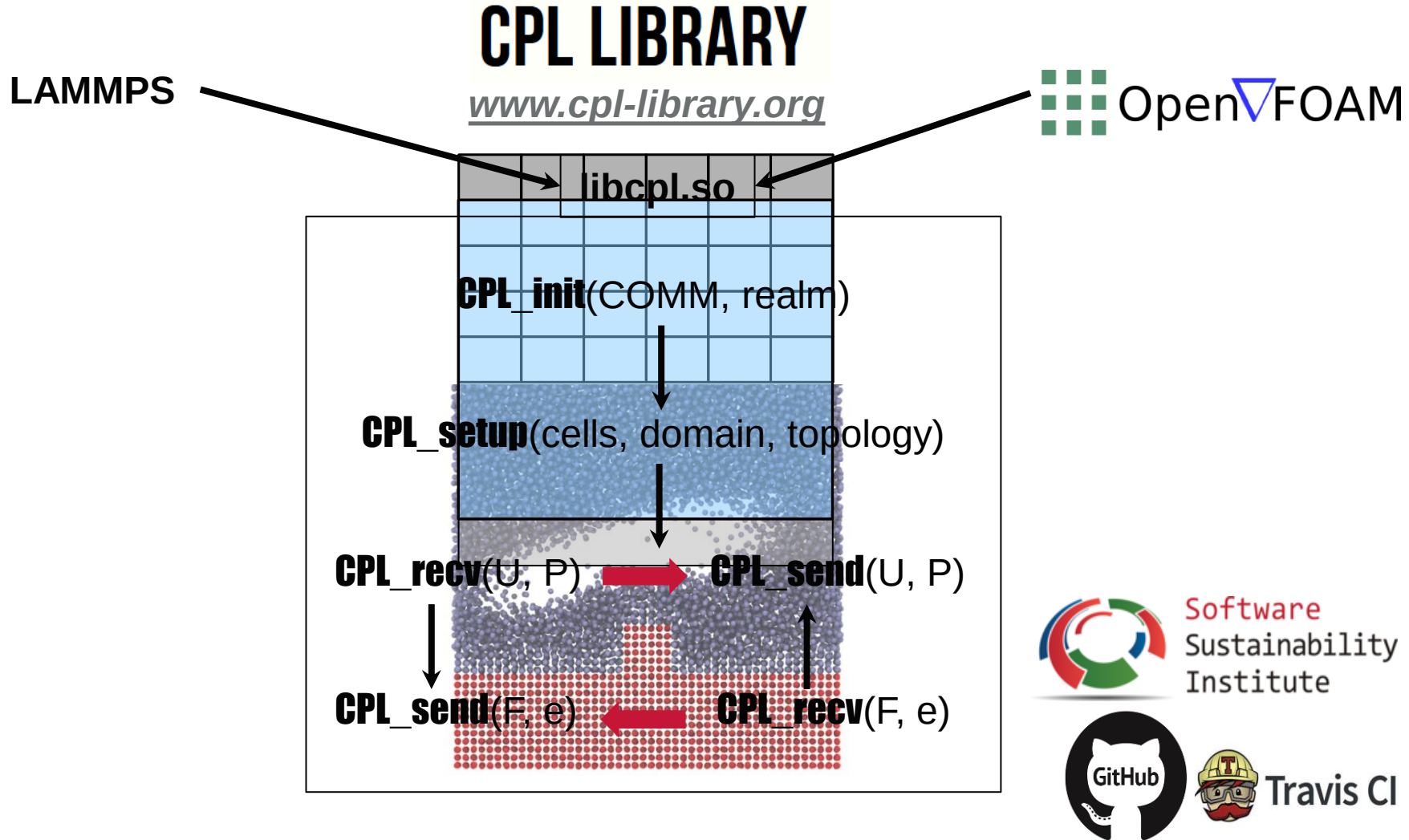


Coupling Results – Couette Flow



Posts shift zero
location





Summary

Control volume for intrinsic surface

$$\vartheta = \int_{V(t)} \delta(\mathbf{r} - \mathbf{r}_i) dV$$

gets $\tilde{\Pi}, \tilde{J}_q$ and γ

Volume moving with fitted spectral surface to molecular spacing

Control volume operator

$$\vartheta = \int_V \delta(\mathbf{r} - \mathbf{r}_i) dV$$

gets Π, J_q, μ and λ

Langevin equation for contact angle

$$\theta^{t+1} = \theta^t - \frac{k\Delta t}{\Gamma} [\theta^t - \langle \theta \rangle] + \xi \frac{\sqrt{C\Delta t}}{\Gamma}$$

